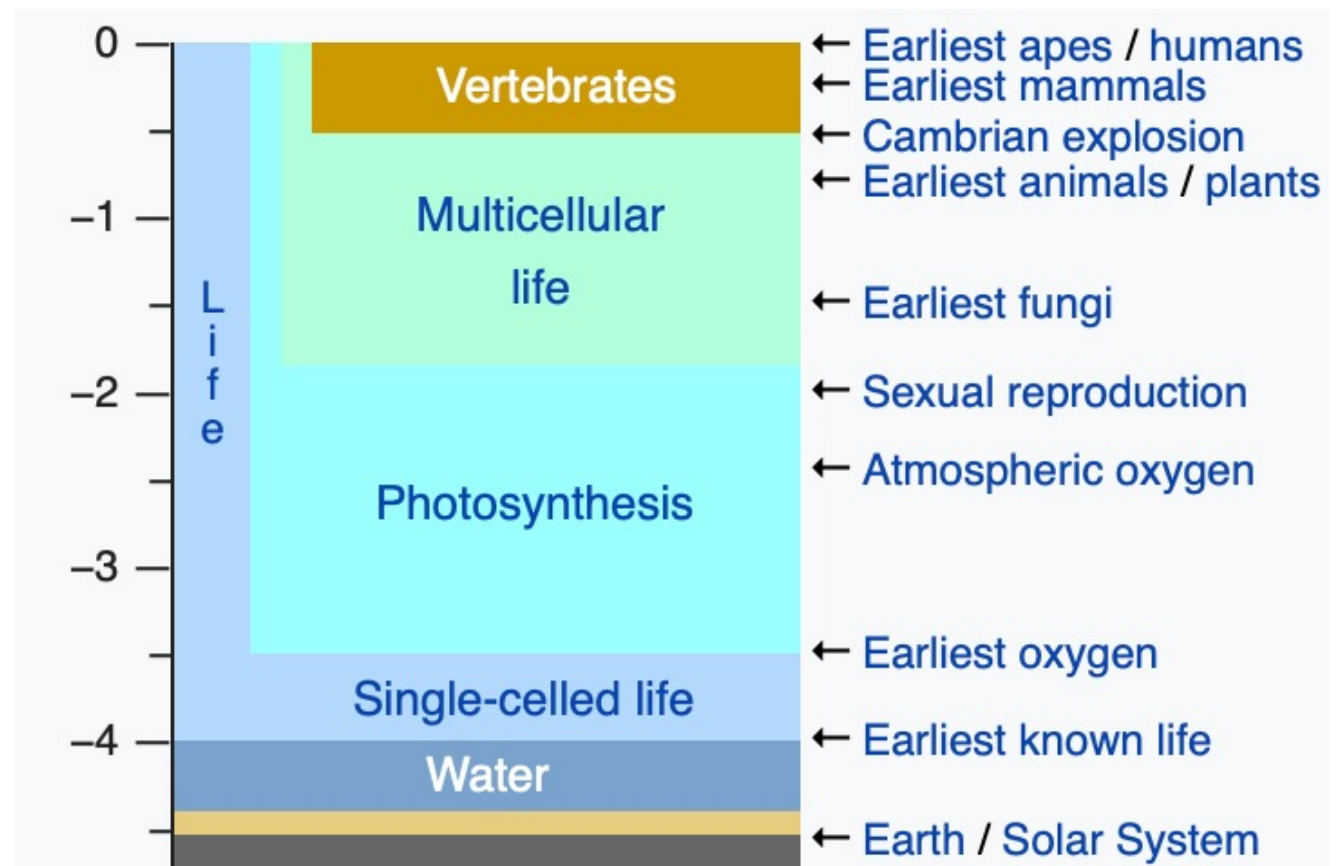


Life

WHAT IS LIFE?

- Organized, contained, chemistry
- Metabolism
 - Obtains & uses energy
 - Obtains & uses materials
- Information
 - Repair
 - Replication
 - Inheritance
 - Evolution



Surface or Deep?

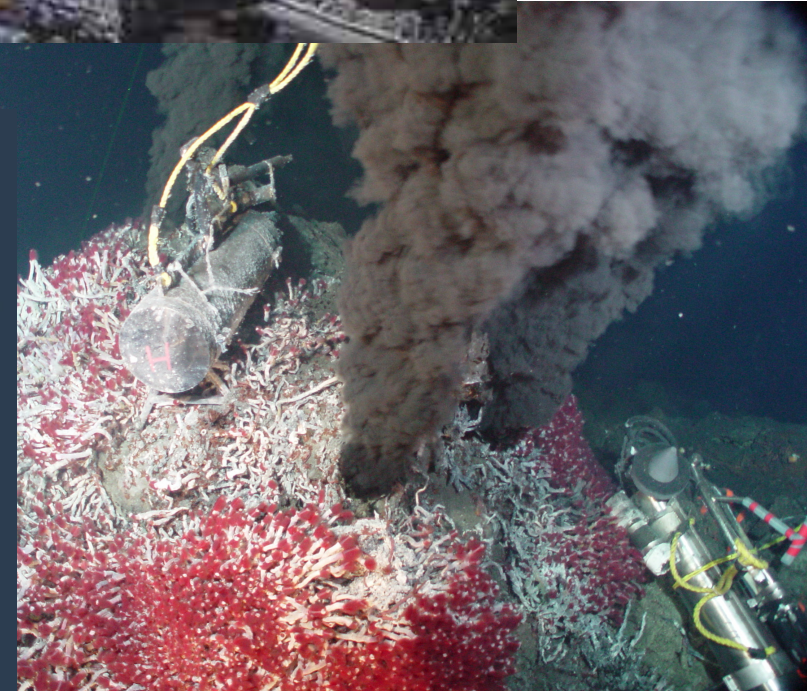
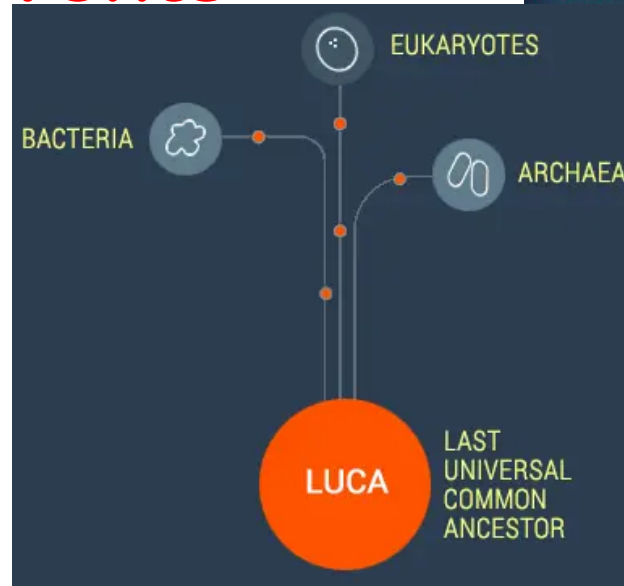
- **Darwin's "warm little pond"**

- Sunlight, UV, lightning
- Miller-Urey experiments
- Tides, *evaporation, concentration*
- Favored by chemists



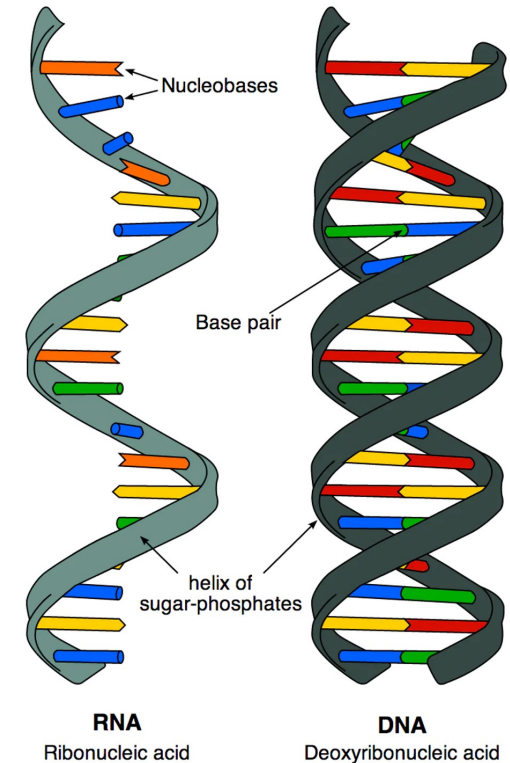
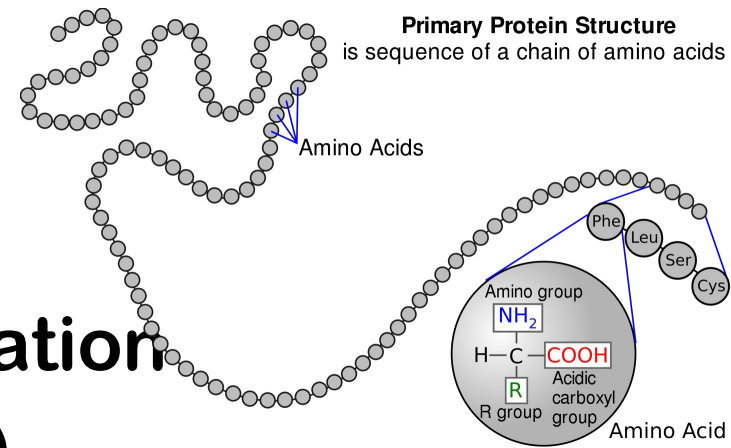
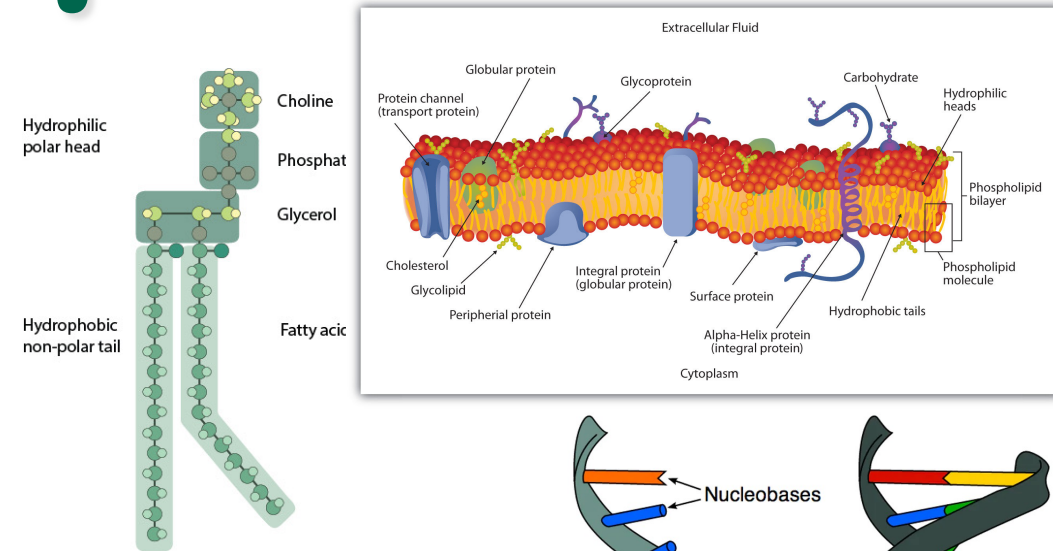
- **Deep sea hydrothermal vents**

- Chemiosmotic energy
- *Genomics* of Last Universal Common Ancestor (LUCA)
- Mineral crystal *chirality*
- Favored by geologists



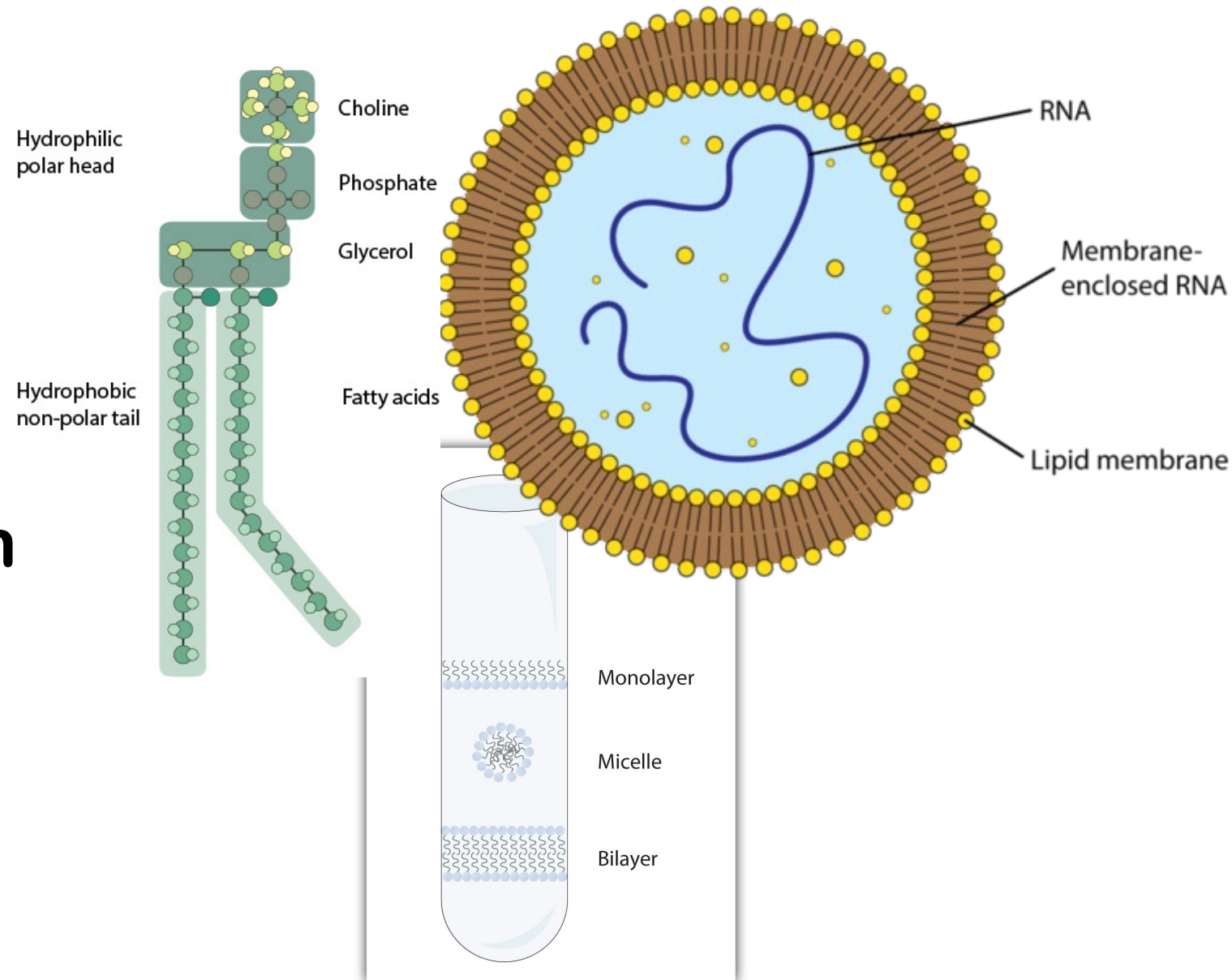
Critical Components

- **CHONPS**
- **Water** (all biochemistry is aqueous)
- **Lipids** (oils, fats for encapsulation)
- **Proteins** (machinery!)
 - Chains, rings, sheets, tubes
 - 20 amino acids (NITROGEN!)
- **Nucleic acids** carry information
 - RNA (codes amino acids in order)
 - DNA (double-helix: makes copies!)



Encapsulation & Membranes

- **Lipids** are hydrocarbons with a water-soluble (polar) and a water-insoluble (hydrophobic) end
- **Molecules line up** when solution dries out
- **Lipid bilayers divide the inside from the outside**



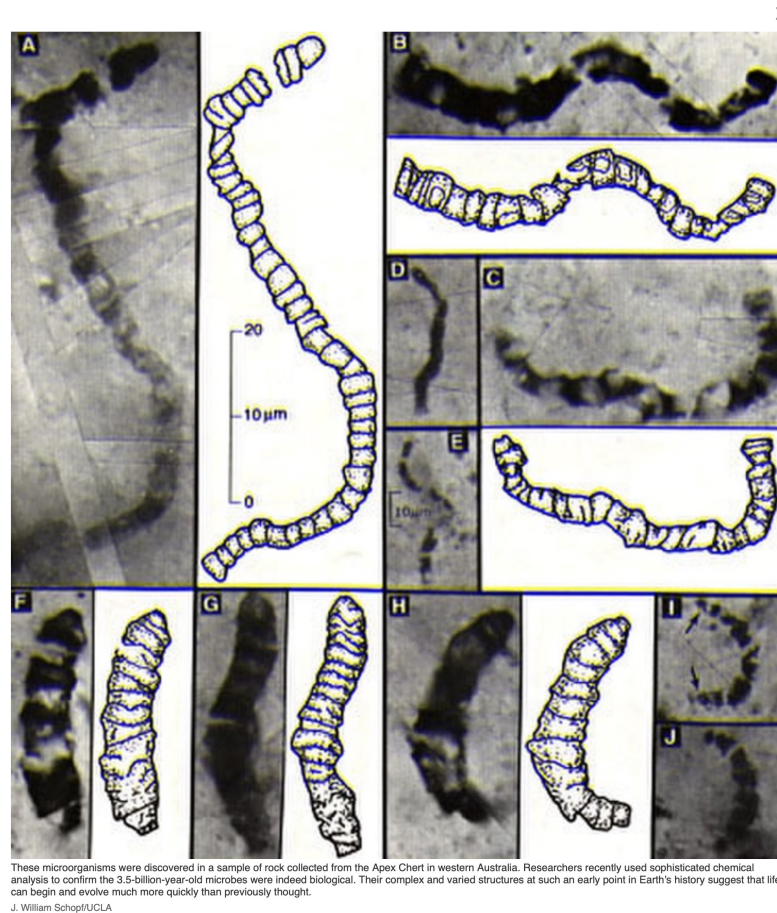
How Old?

3.5 billion years?

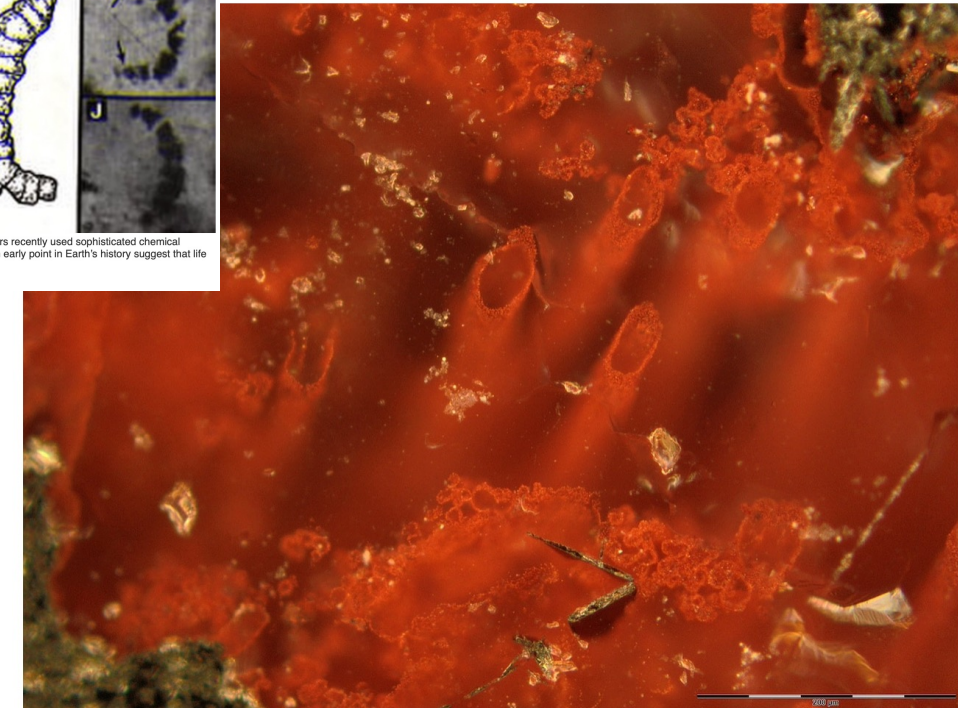
- Apex chert western Australia
- 11 complex microbes of 5 distinct species
- Depleted in ^{13}C

4.28 billion years?

- Hematite rods from undersea vents in Quebec
- Before Late Heavy Bombardment!



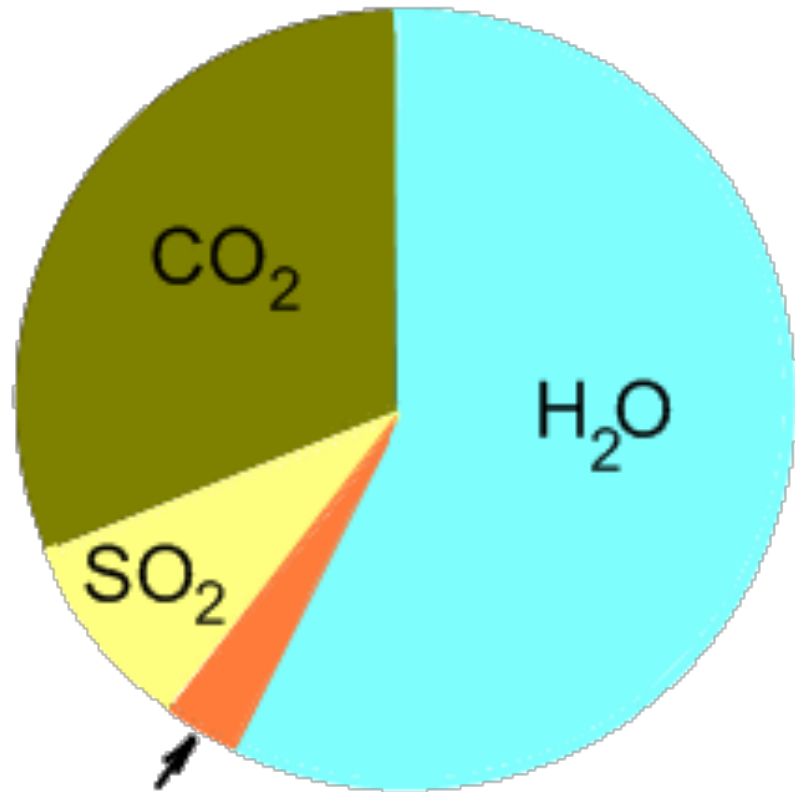
These microorganisms were discovered in a sample of rock collected from the Apex Chert in western Australia. Researchers recently used sophisticated chemical analysis to confirm the 3.5-billion-year-old microbes were indeed biological. Their complex and varied structures at such an early point in Earth's history suggest that life can begin and evolve much more quickly than previously thought.
J. William Schopf/UCLA



Seen under a microscope, these iron-rich tubes may be the oldest known fossils on the planet.
PHOTOGRAPH COURTESY MATTHEW DODD

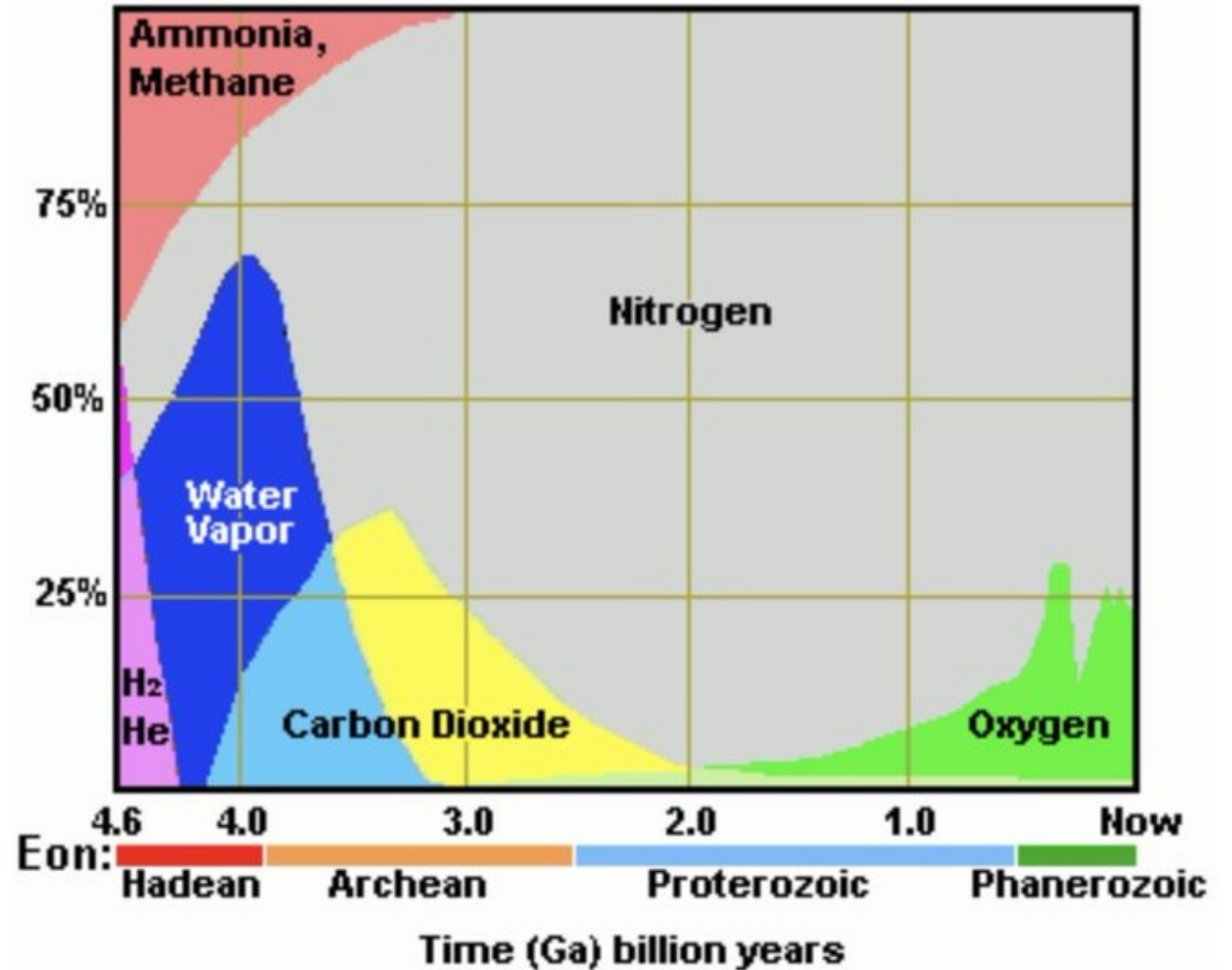
Evolution of the Atmosphere

Volcanic Gases



All Remaining Volcanic Gases

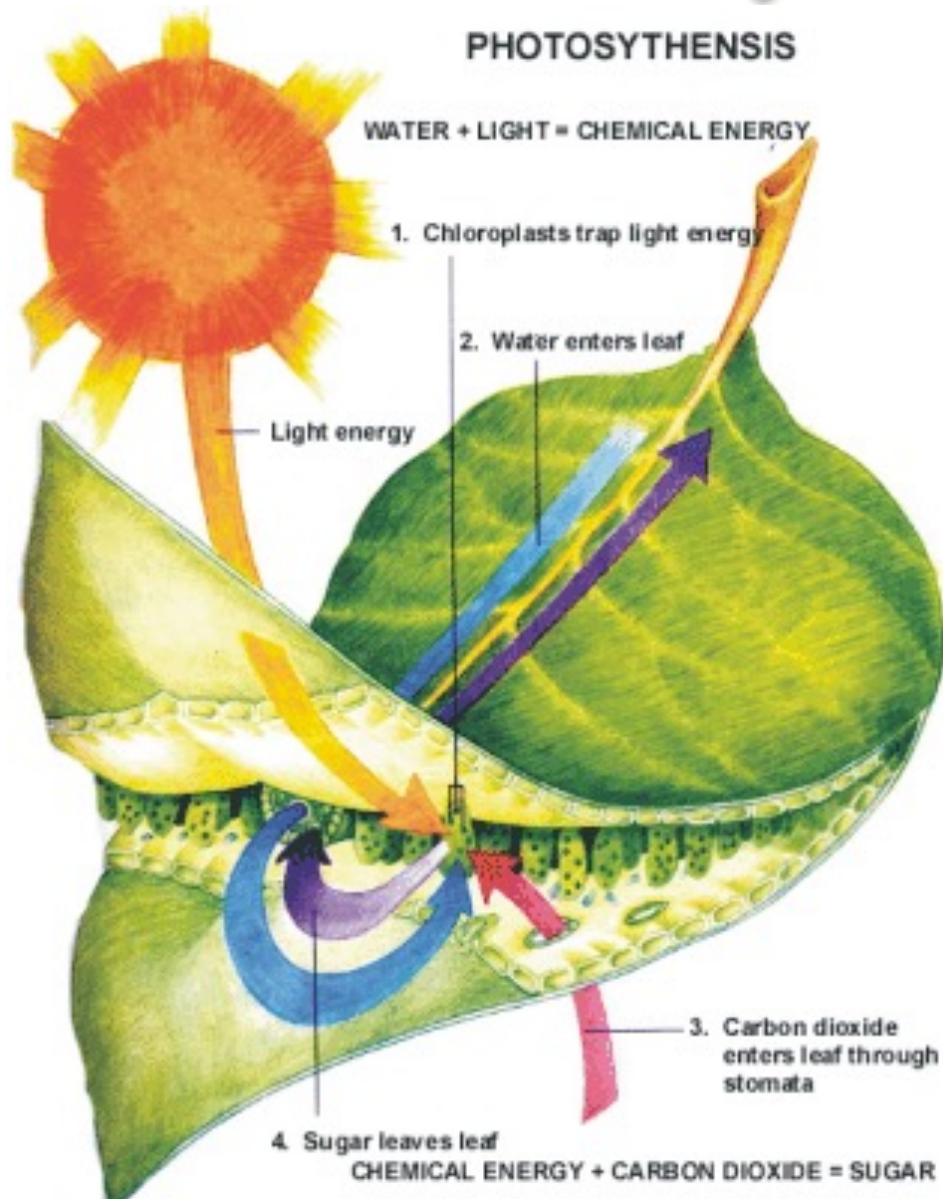
% of Atmosphere Composition of Earth's atmosphere



Oxygen Doesn't Want to be Free!

- It takes a *lot* of energy to free oxygen
- Energy is liberated when oxygen is consumed
- Photosynthesis is energy & chemical source of free oxygen, but ...
- net O₂ production requires mummification/burial/fossilization of organic matter to prevent respiration
- Meanwhile, O₂ is consumed by weathering & volcanic gases

Carbon, Life, & Energy



- Photosynthesis uses energy from the sun to **convert inorganic air (CO_2) to living biomass!**
- Almost all of this energy is **released through respiration (back to CO_2)** when plants are eaten by animals, bacteria, people
- **$1/7$ of all CO_2 every year!**

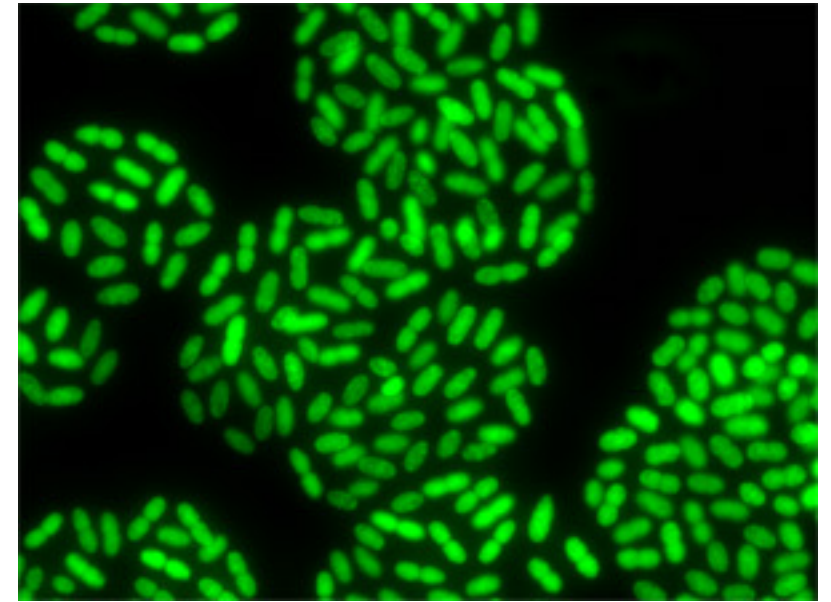


“IT’S ALIVE!”



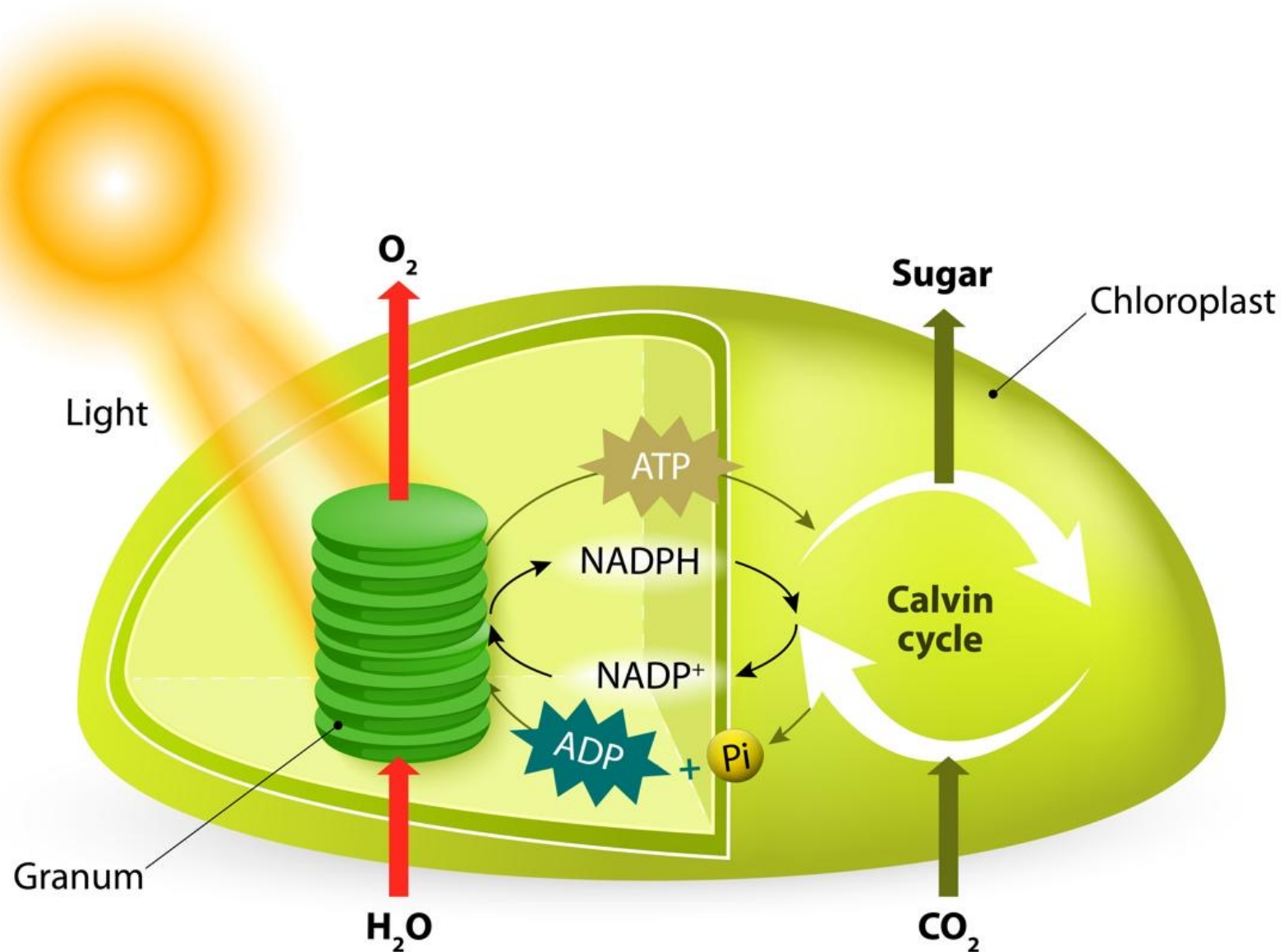
PHOTOSYNTHESIS

Cyanobacteria

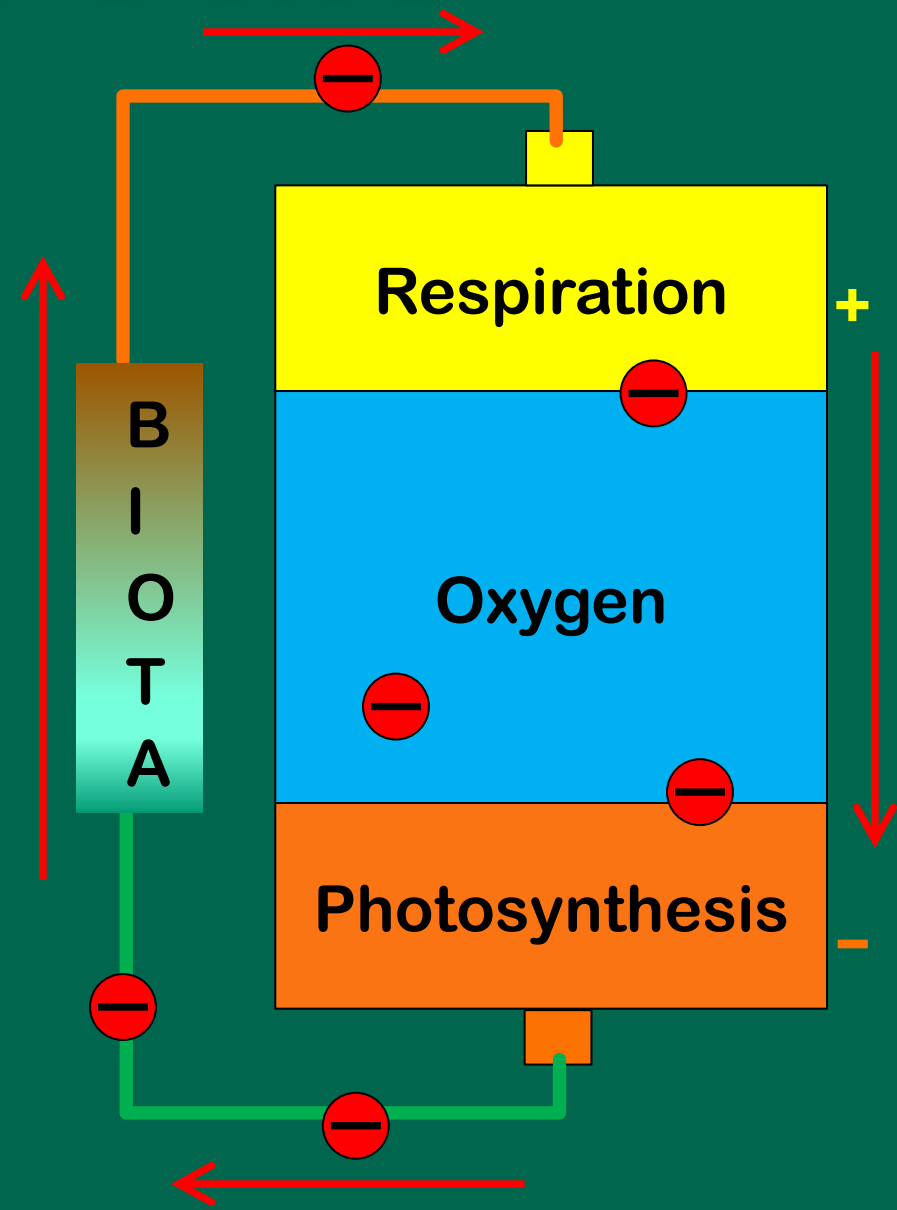
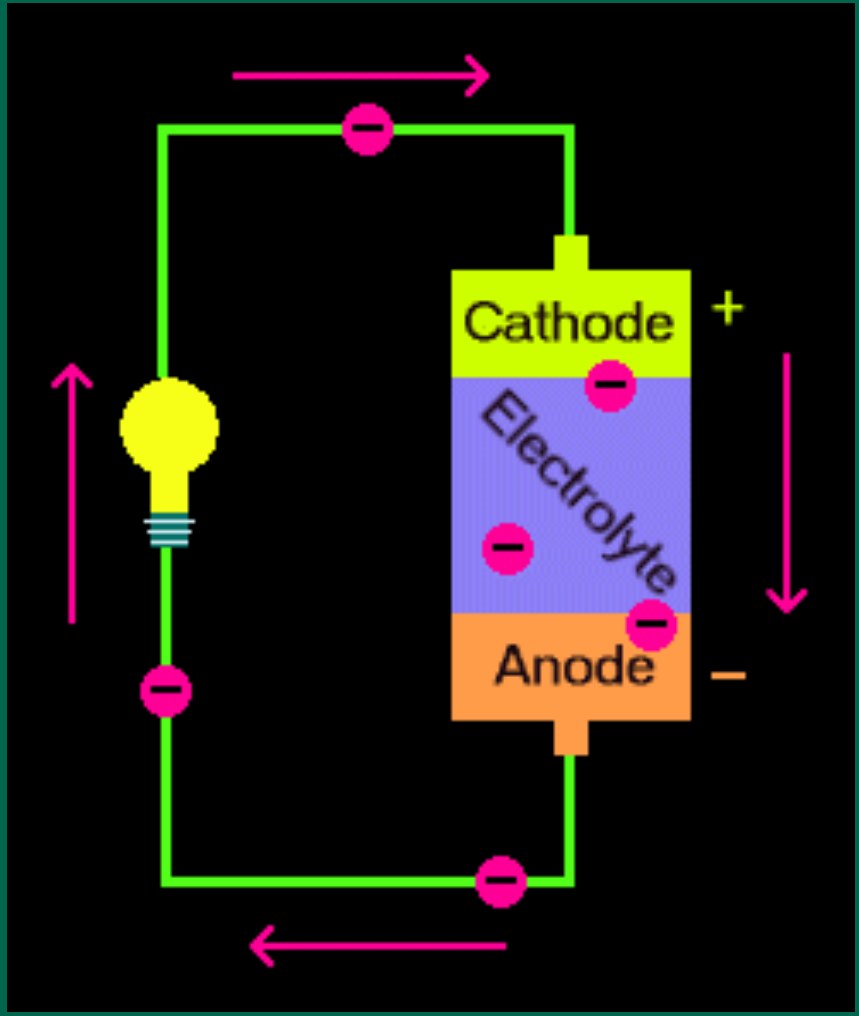


3.5 billion yr

PROCESS OF PHOTOSYNTHESIS

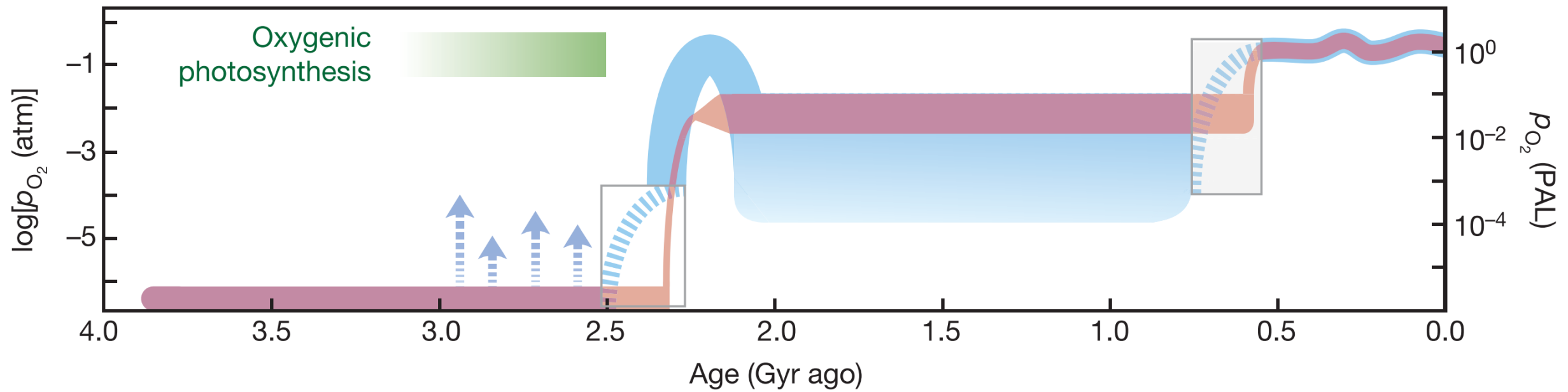


Life is a Circuit



Oxygenation

Faint Young Sun Paradox Snowball Earth “Boring Billion” Cambrian Explosion



Modern Stromatolites

layered
microbial
mats

Shark Bay
Western
Australia



A photograph of Prof. Don Boyd, a stromatolite expert, in a field of stromatolites. He is wearing a light-colored bucket hat, a blue denim jacket, blue jeans, and yellow gloves. He is holding a blue measuring tool (a dipper or similar) and pointing it towards a large, layered rock formation. The background shows a rocky hillside with sparse vegetation under a clear blue sky. The text "Prof. Don Boyd Stromatolite Expert" is overlaid in red on a semi-transparent white box in the upper right corner.

**Prof. Don Boyd
Stromatolite Expert**



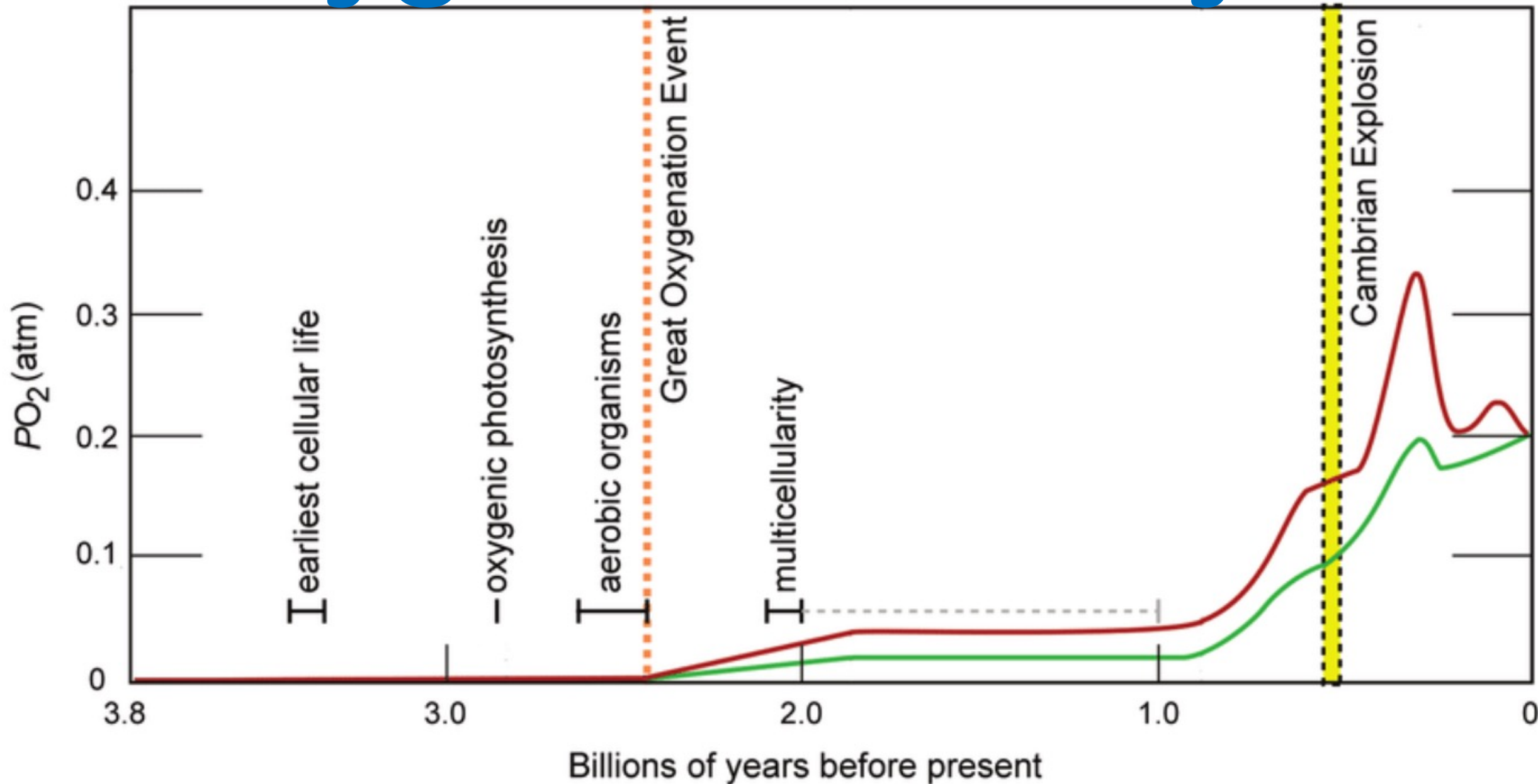








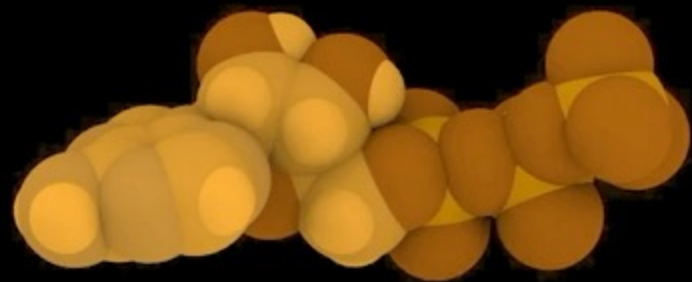
Oxygen! Run Away!



Respiration

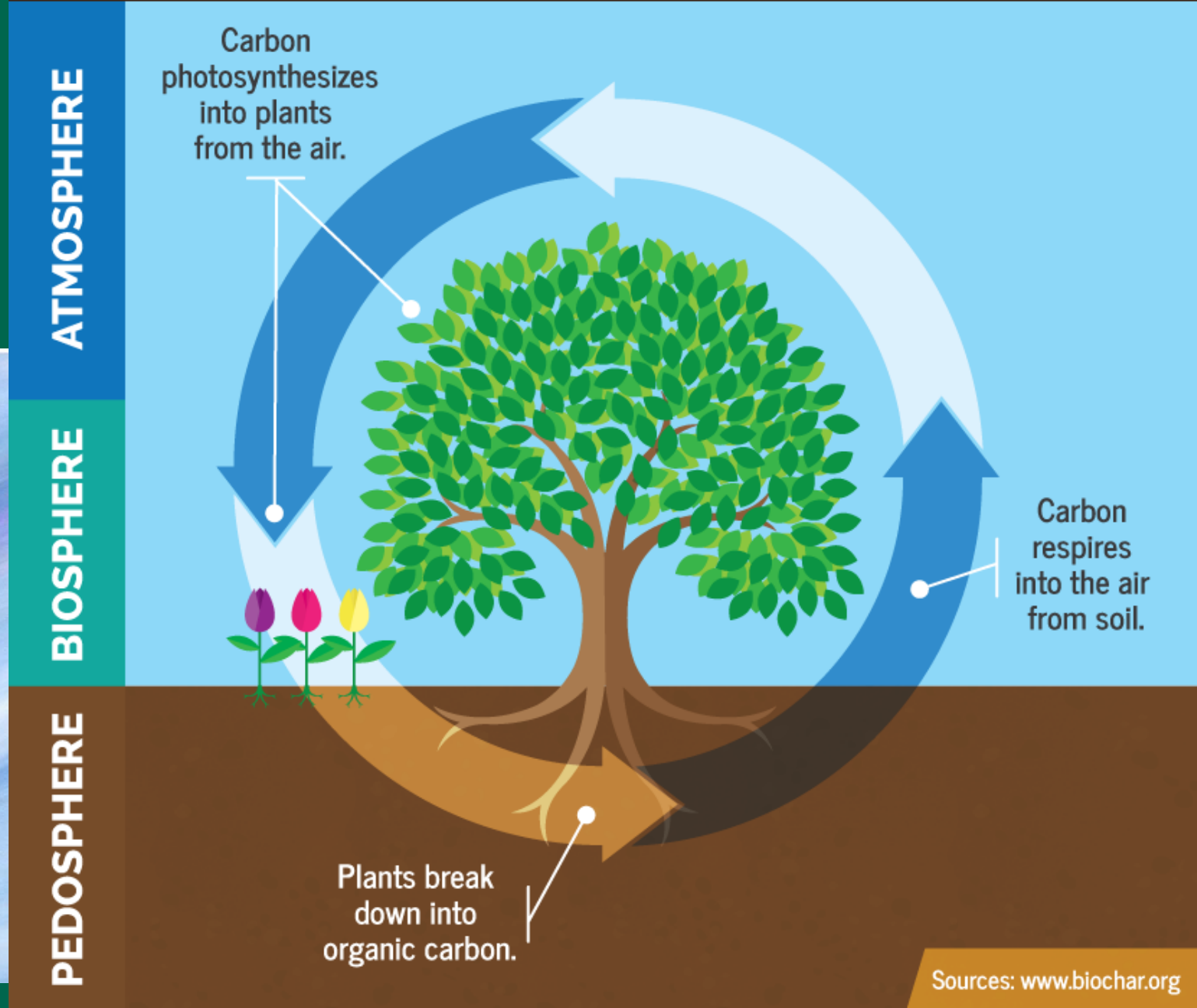
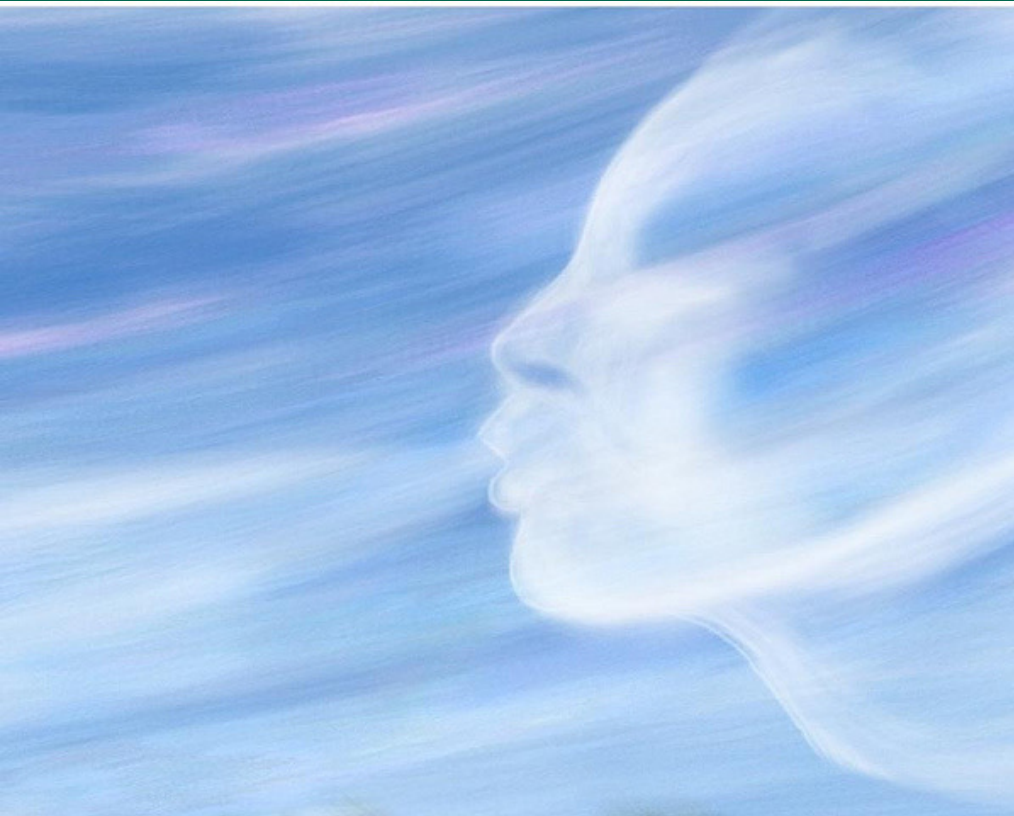
The flame that releases the stored energy of the Sun, harvesting the organic stuff of which we're made





Breathing of the Earth

THE CARBON CYCLE



T

s

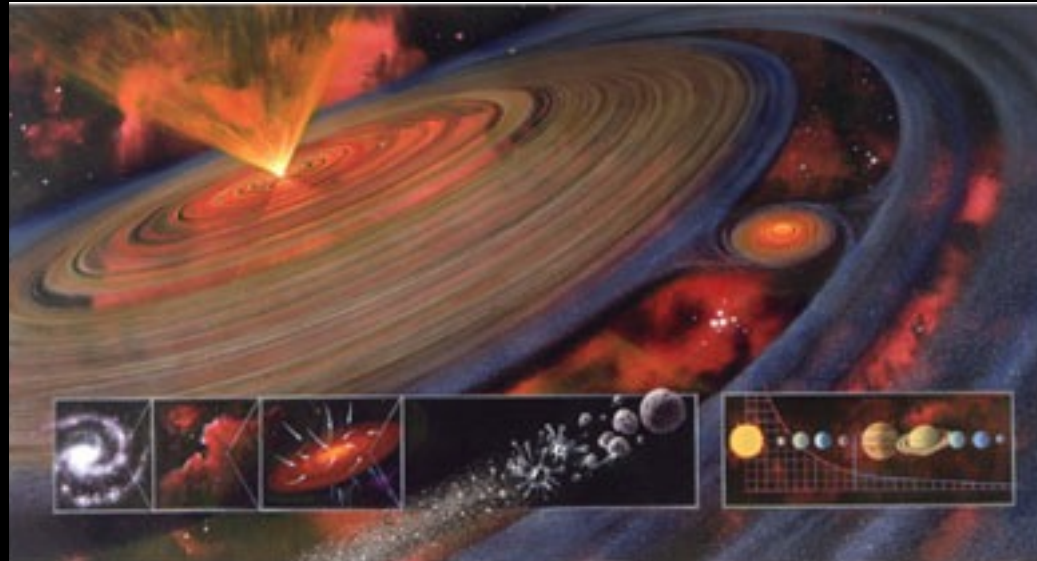
-

h

The Solar System

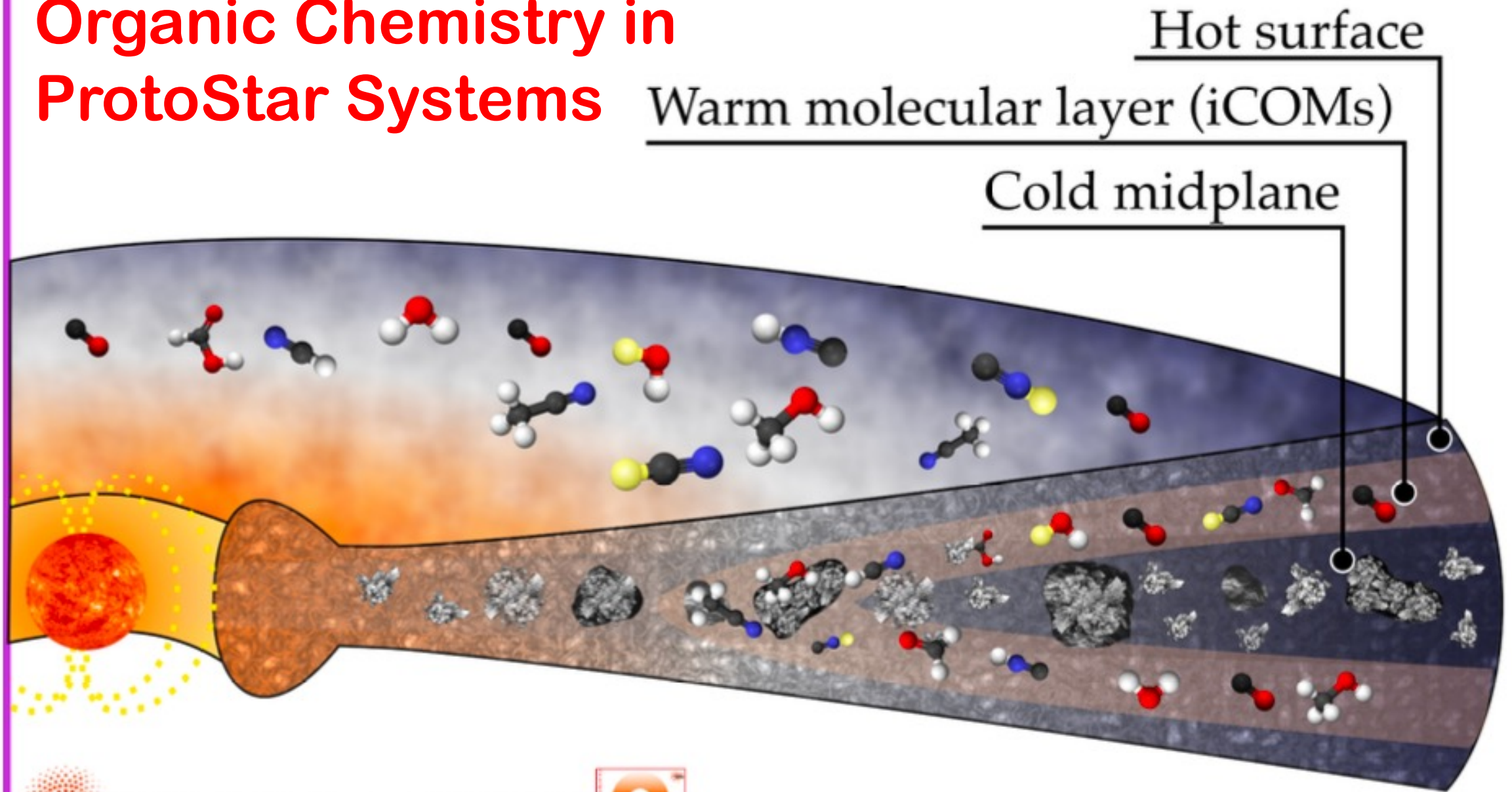
Solar System Origins

- Gravitational collapse
- Self-gravitating Hydrogen
- Swirling dust of a dozen Ur-Minerals



- Volatiles: wet and dry ice beyond the “snowline”
- Heating, hydration, & oxidation of Ur-Minerals to form about 50 primordial minerals
- Accretion, collisions, & vertical differentiation
- Metallic cores, rocky mantles, organic gunk

Organic Chemistry in ProtoStar Systems



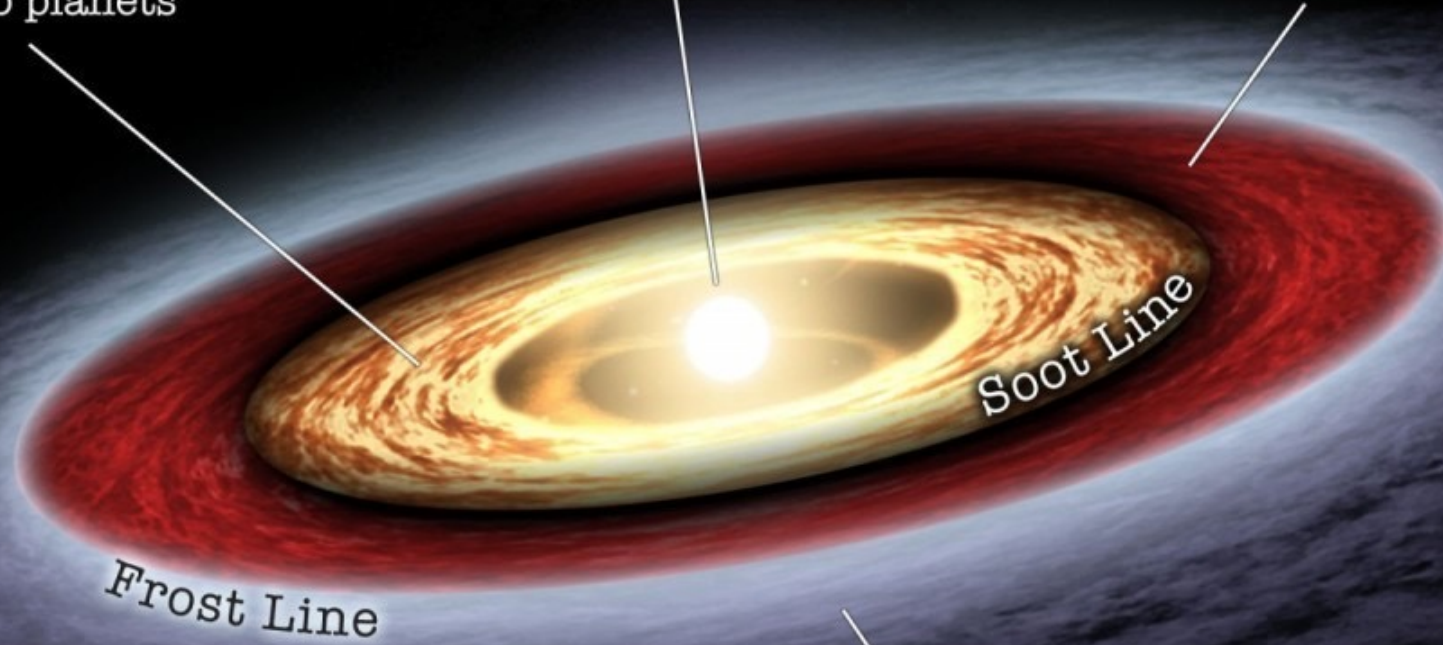
Condensation/Deposition of Dust



Protostar

Central Region –
Only metals and minerals
condense into planets

Outside the Soot Line –
PAHs exist, allowing forming planets to
include condensed carbon compounds



Soot Line

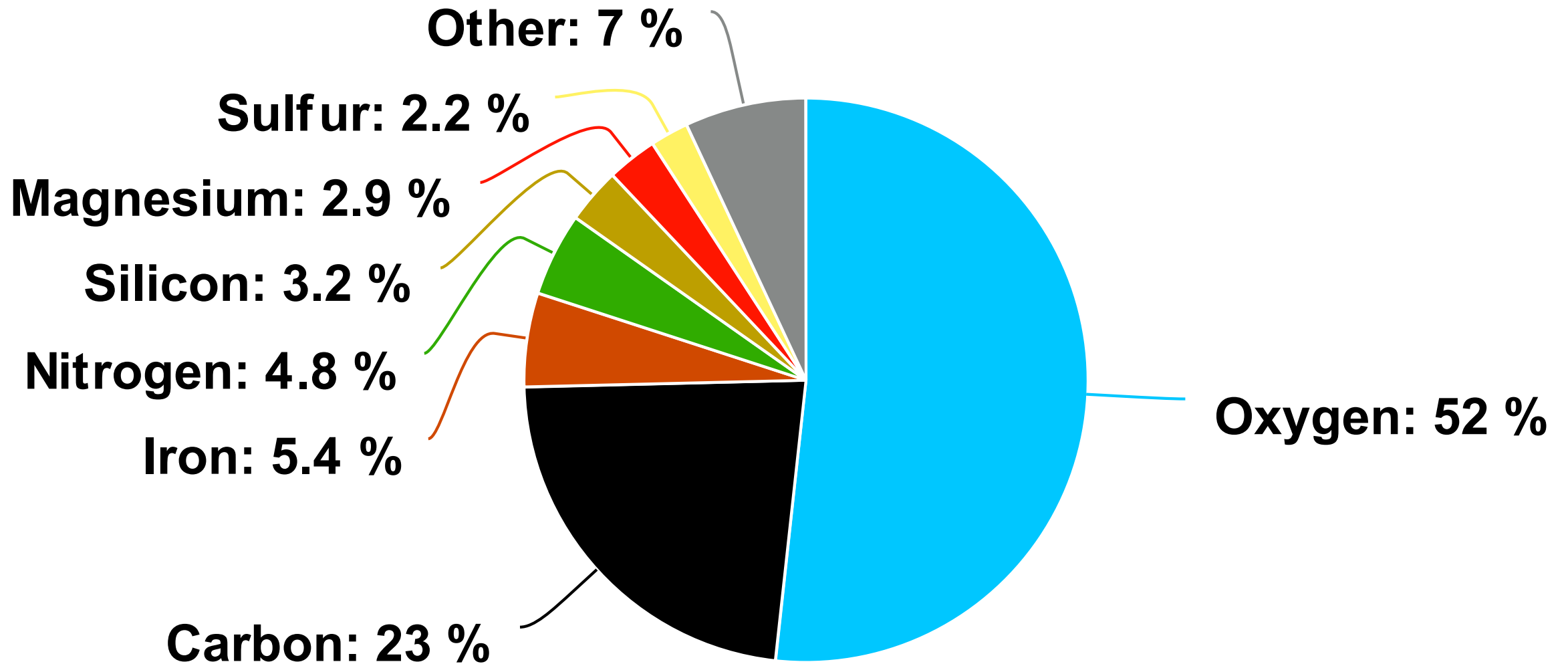
Frost Line

Outside the Frost Line –
Low temperatures allow condensing planets
to include volatile molecules such as H_2O ,
 NH_3 and CH_4

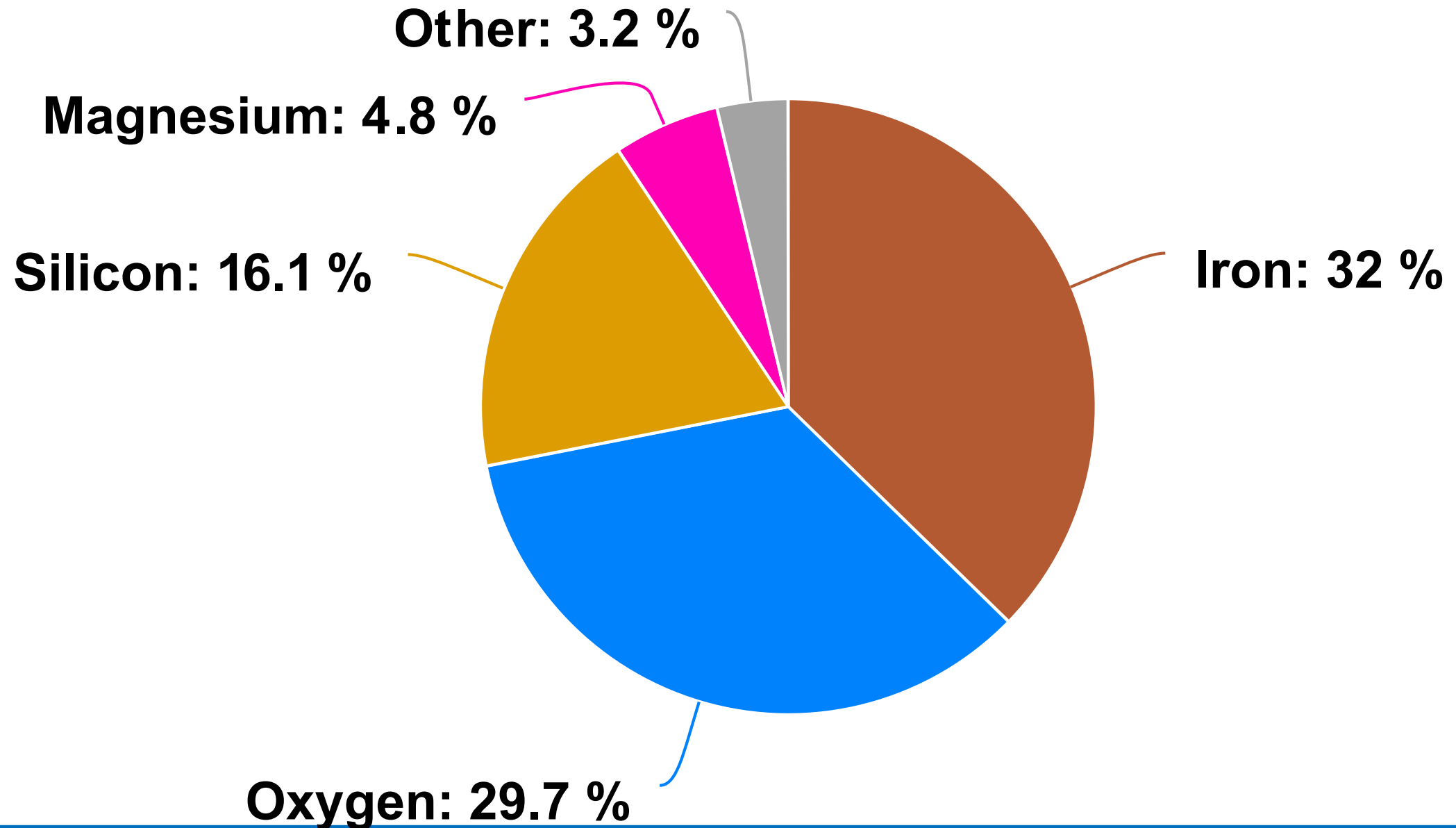
~98% of the nebula is hydrogen and helium
which do not condense

Our Galaxy

beyond H (74%) & He (24%)

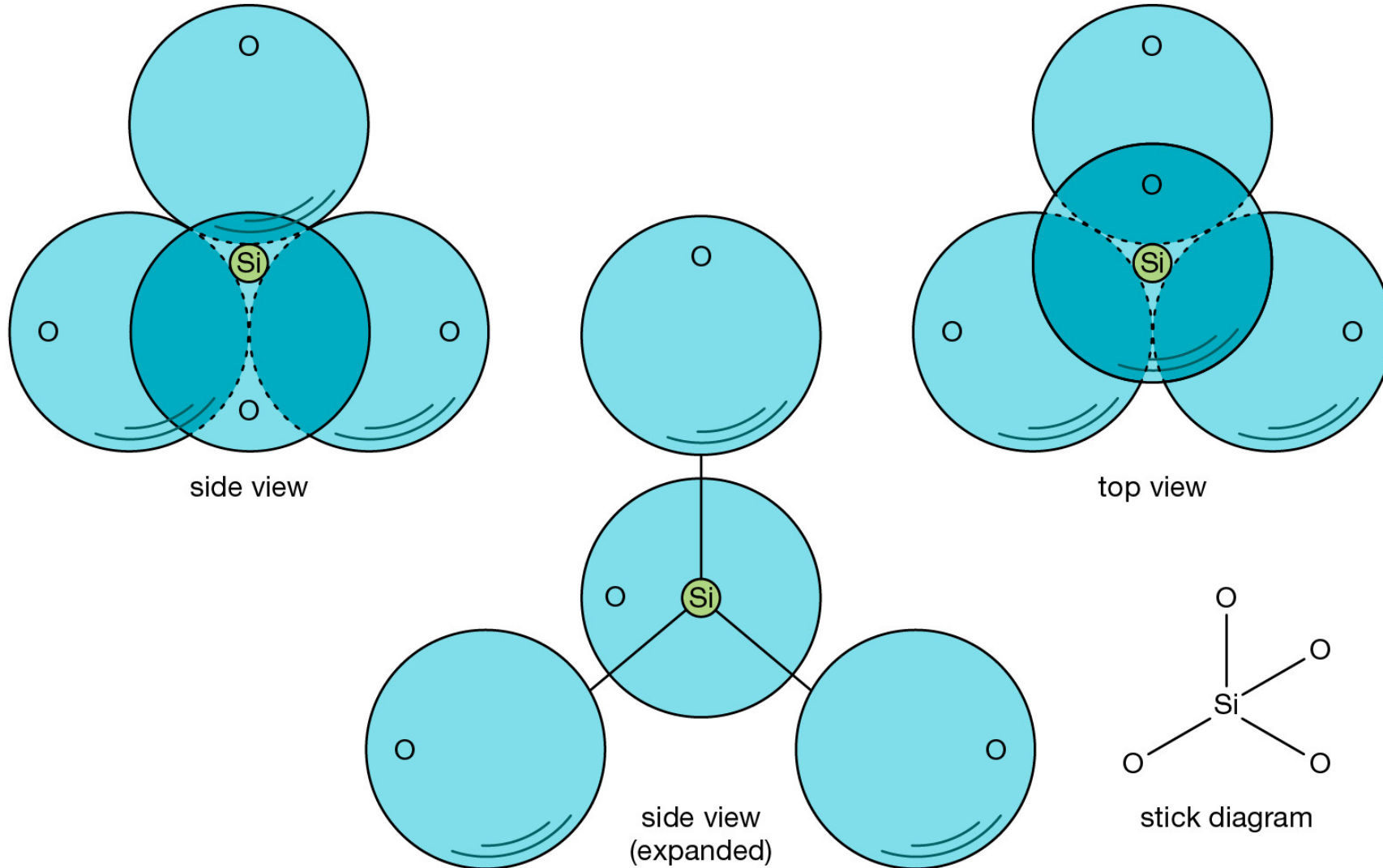


Earth



Building with Pyramids

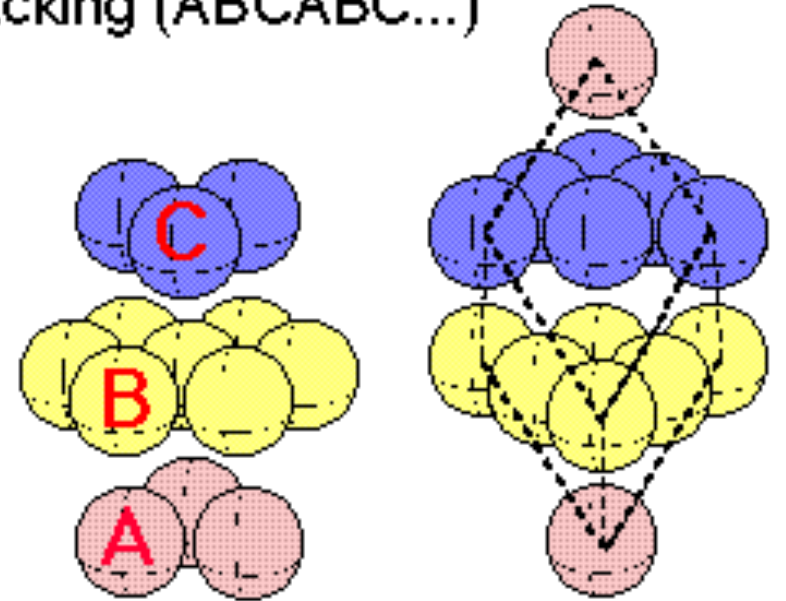
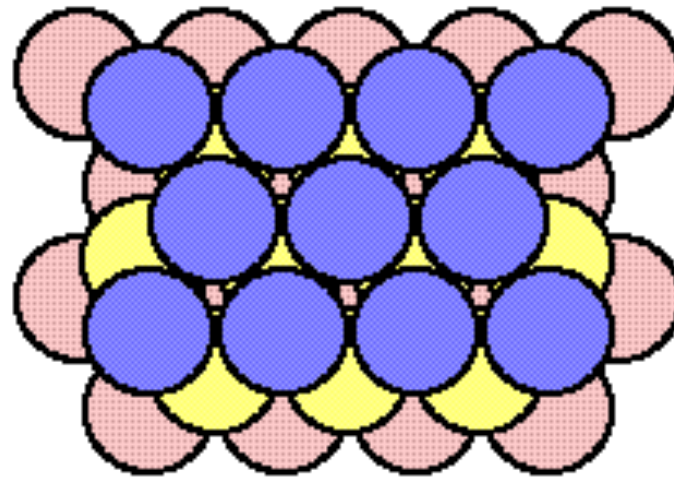
Silicon-oxygen tetrahedron



Rocks = Oxygen Atoms Packed Together!



Cubic close packing (ABCABC...)



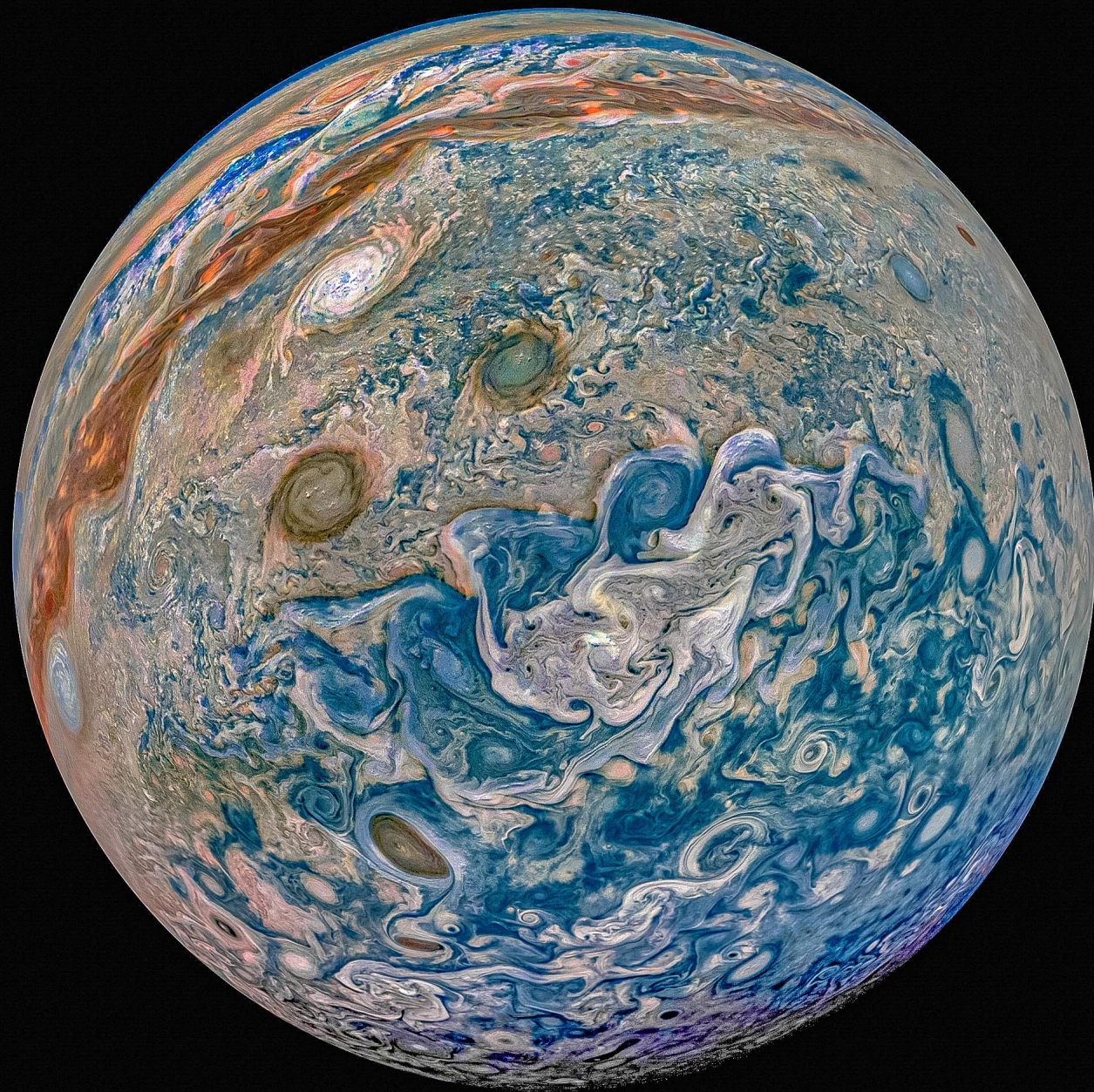
All other elements fit into the tiny void spaces!

Ingredients

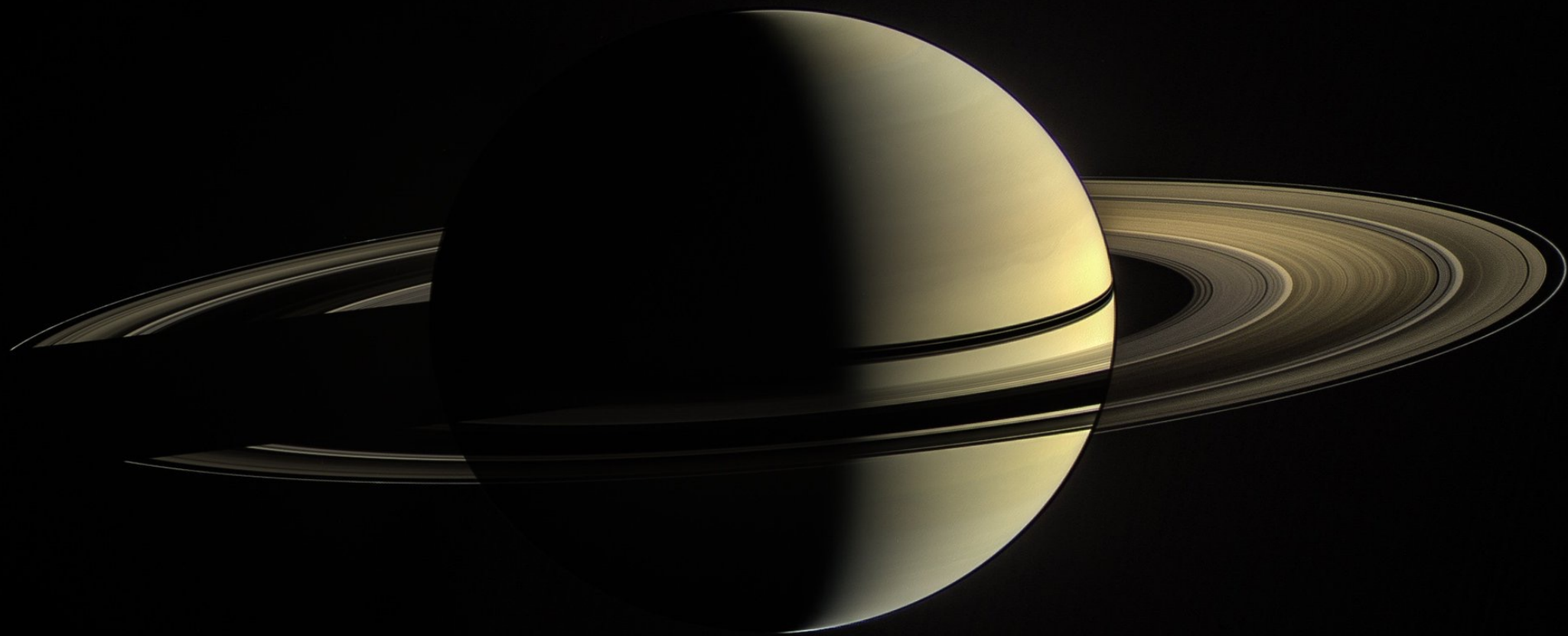
Gas and Ice

Rocks

1 H Hydrogen 1.008																	2 He Helium 4.002602
3 Li Lithium 6.94	4 Be Beryllium 9.012182											5 B Boron 10.81	6 C Carbon 12.011	7 N Nitrogen 14.007	8 O Oxygen 15.999	9 F Fluorine 18.998403163	10 Ne Neon 20.1797
11 Na Sodium 22.98976928	12 Mg Magnesium 24.304	13 Al Aluminium 26.9815385	14 Si Silicon 28.0855	15 P Phosphorus 30.973761508	16 S Sulfur 32.06	17 Cl Chlorine 35.45	18 Ar Argon 39.948										
19 K Potassium 39.0983	20 Ca Calcium 40.078	21 Sc Scandium 44.955908	22 Ti Titanium 47.867	23 V Vanadium 50.9415	24 Cr Chromium 51.9961	25 Mn Manganese 54.938044	26 Fe Iron 55.845	27 Co Cobalt 58.933194	28 Ni Nickel 58.6934	29 Cu Copper 63.546	30 Zn Zinc 65.38	31 Ga Gallium 69.723	32 Ge Germanium 72.630	33 As Arsenic 74.921595	34 Se Selenium 78.971	35 Br Bromine 79.904	36 Kr Krypton 83.798

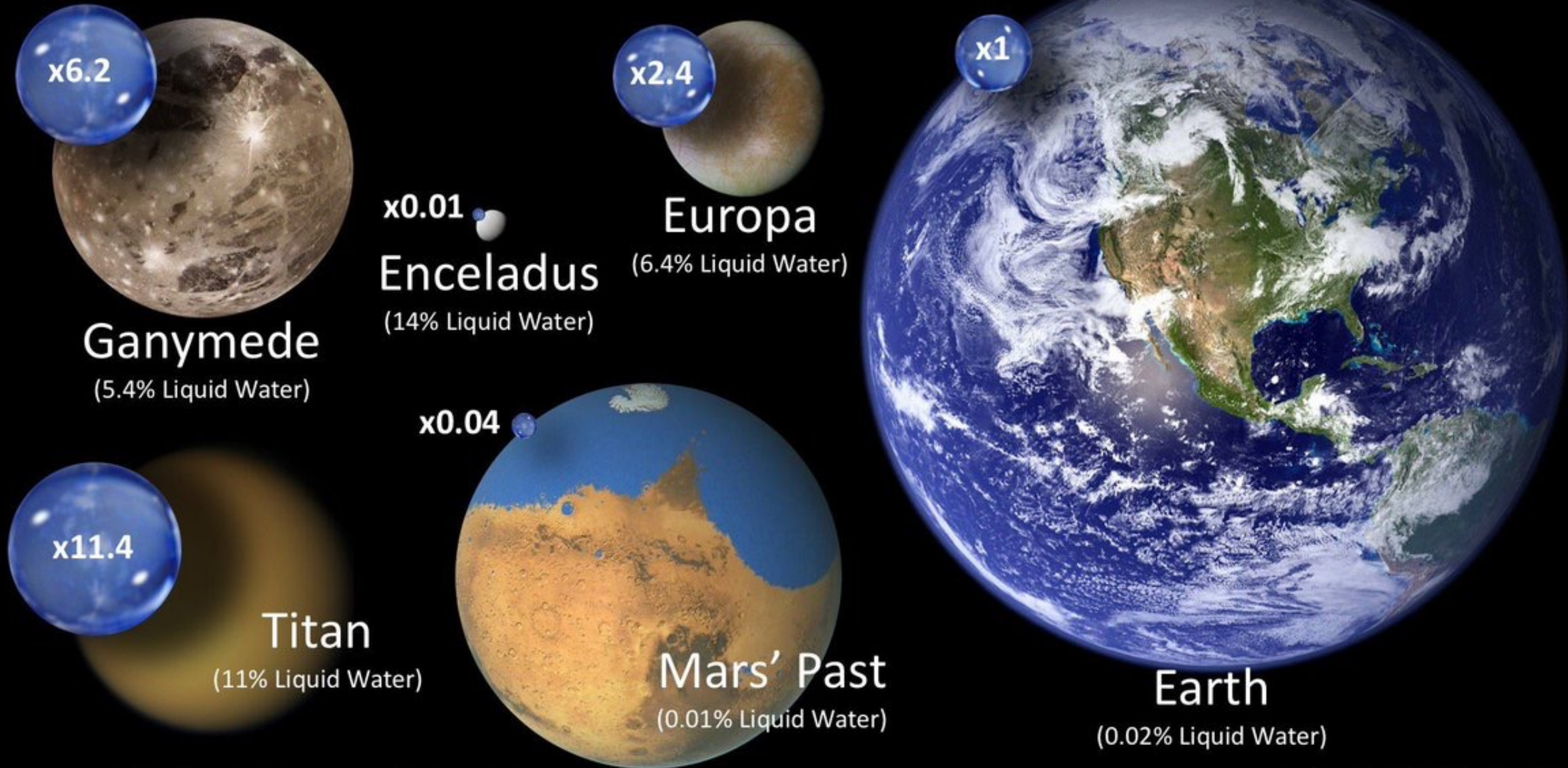








Oceans in the Solar System

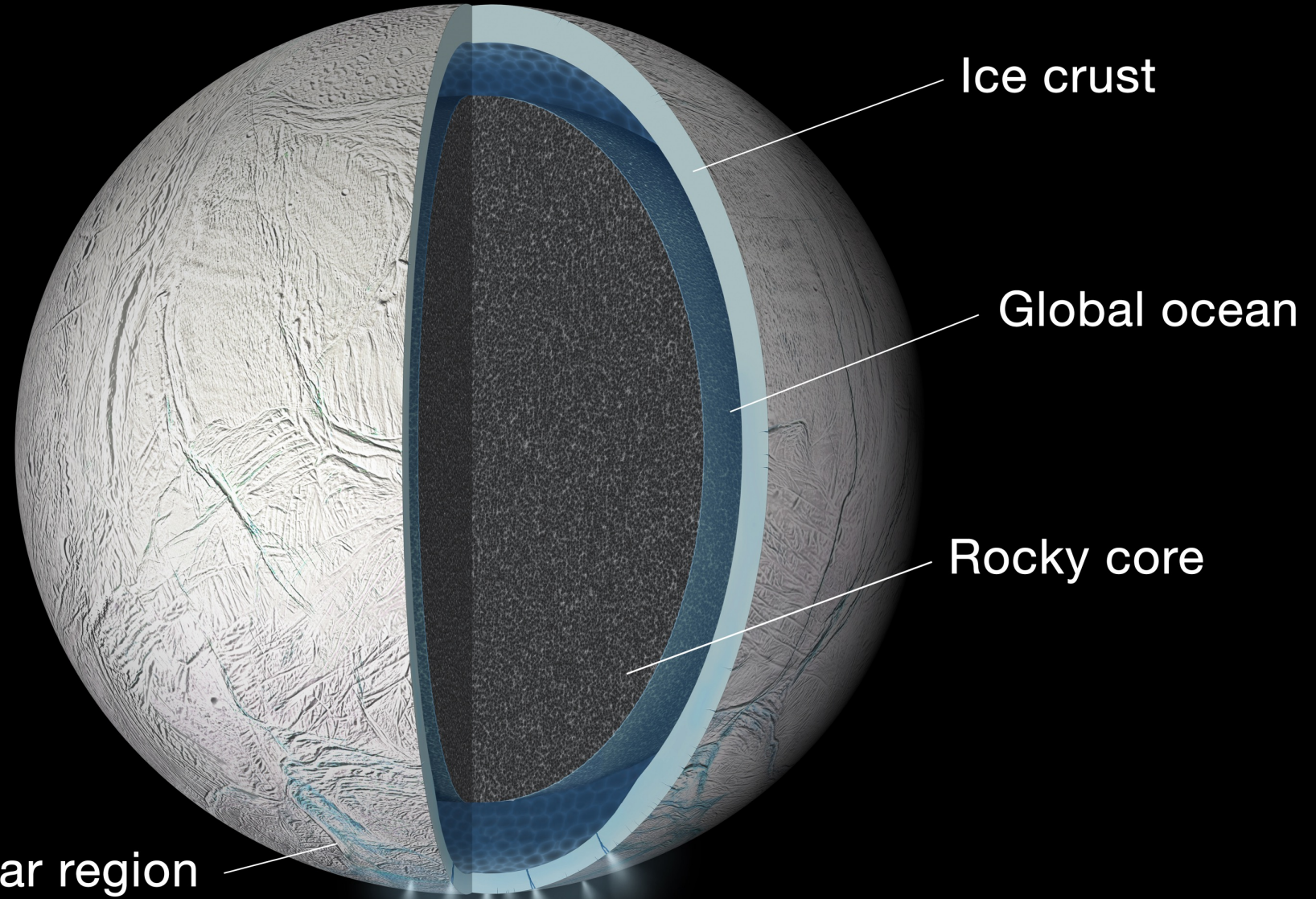


(mass percent of liquid water between parenthesis, excluding water ice)

Credit: PHL @ UPR Arcibo, NASA

Global Ocean on
Saturn's Moon
ENCELADUS

South polar region
with active jets

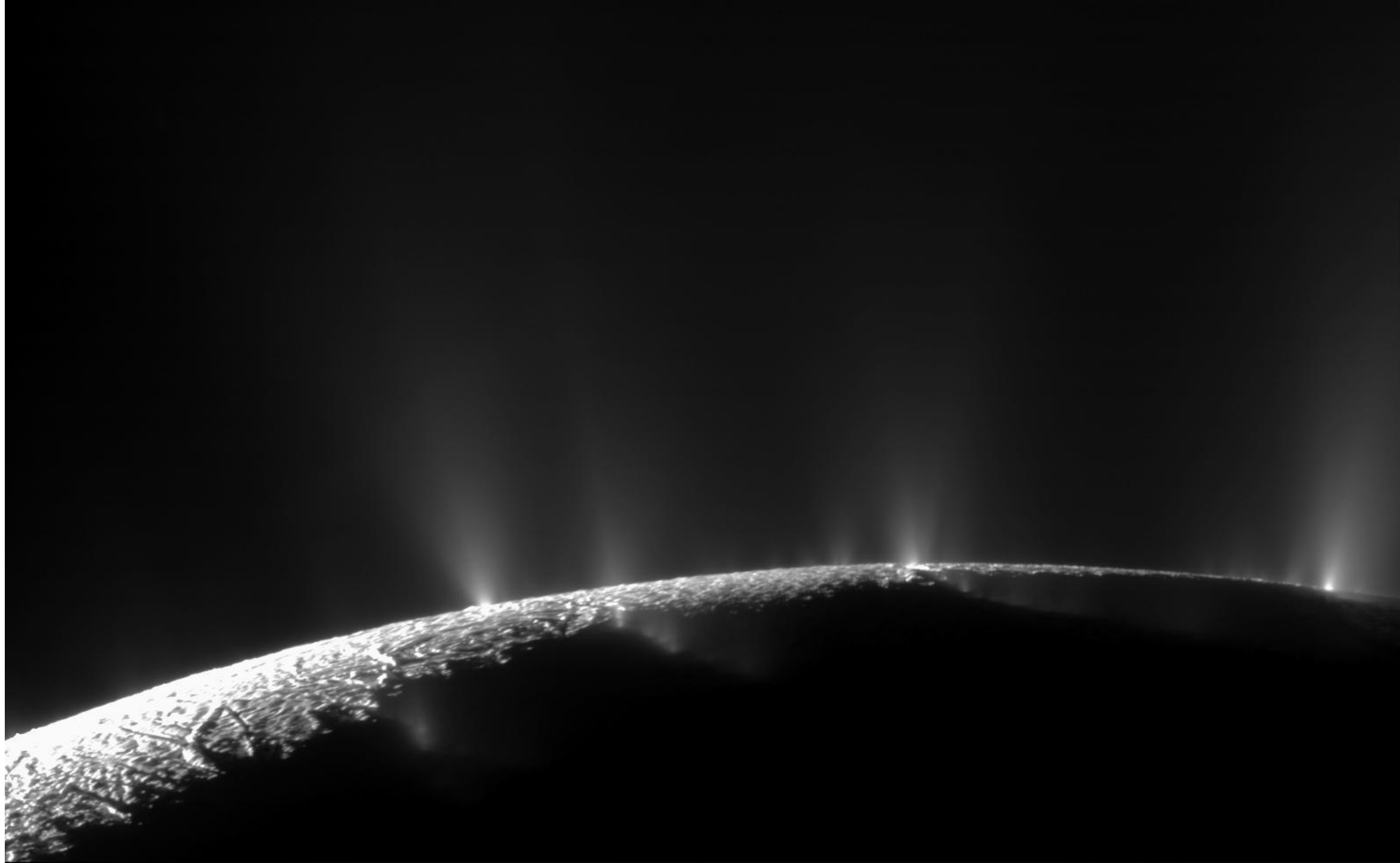


Ice crust

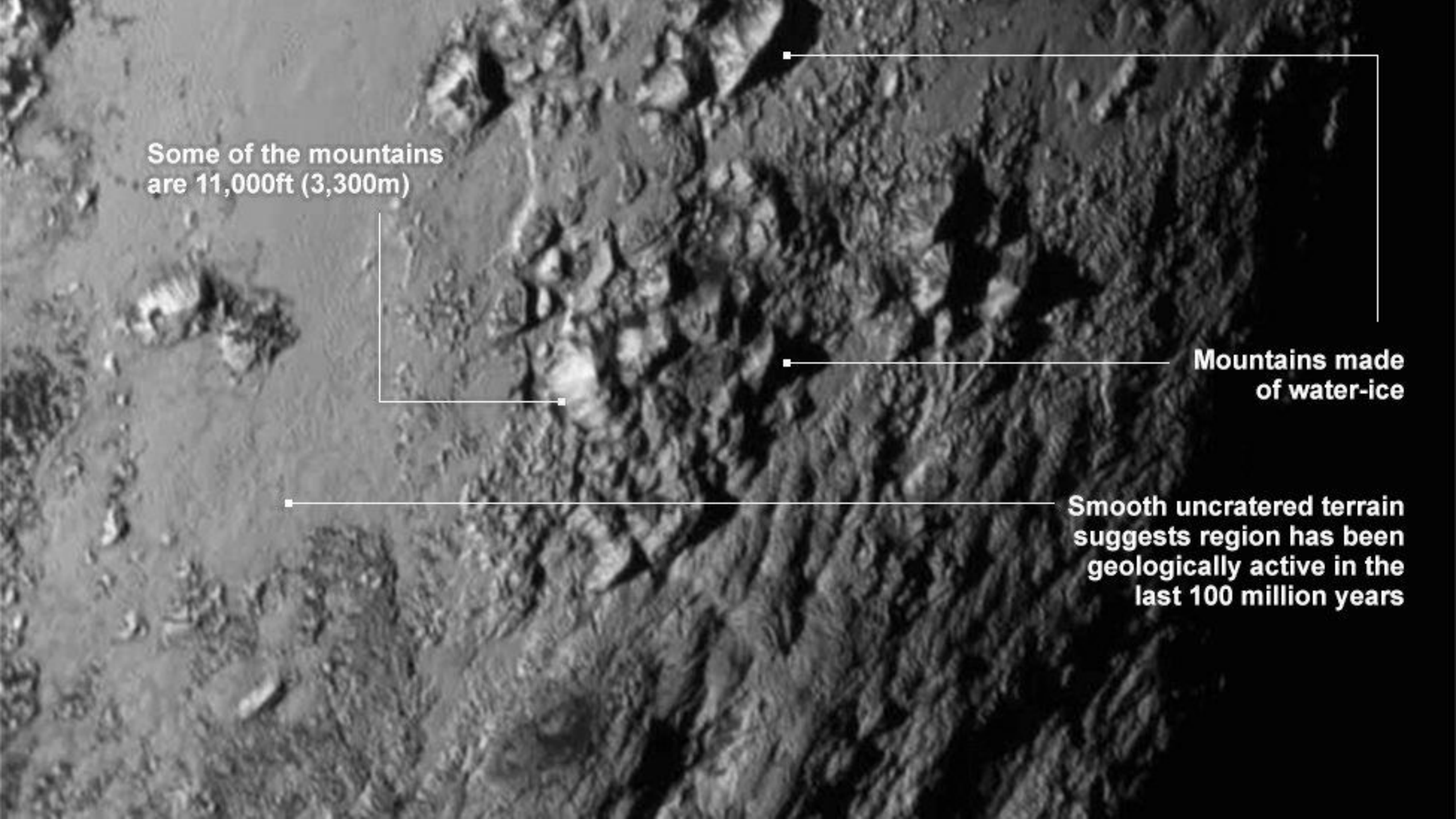
Global ocean

Rocky core

* Thickness of layers is not to scale





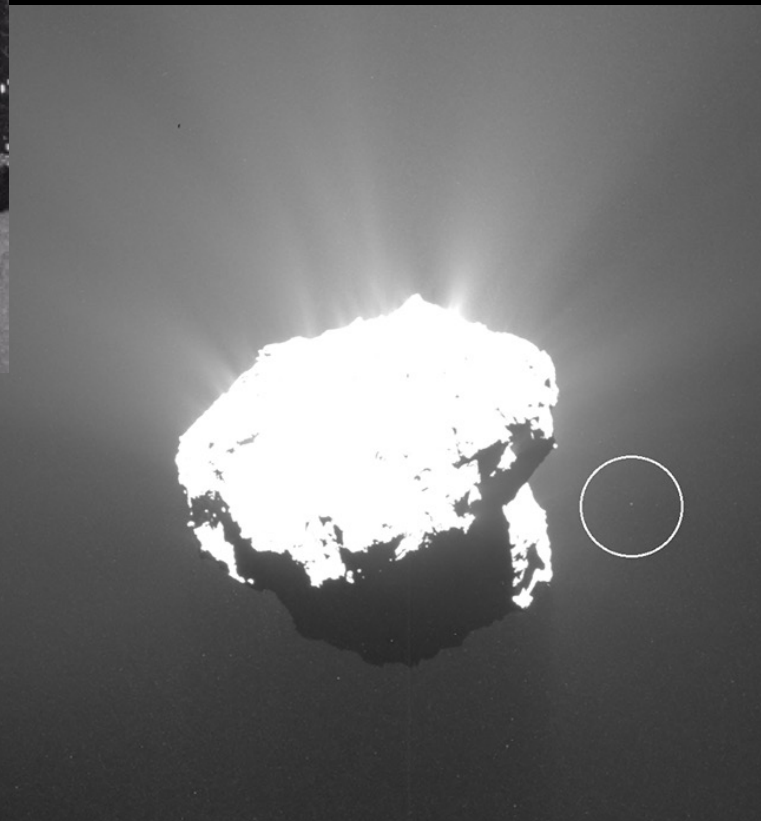
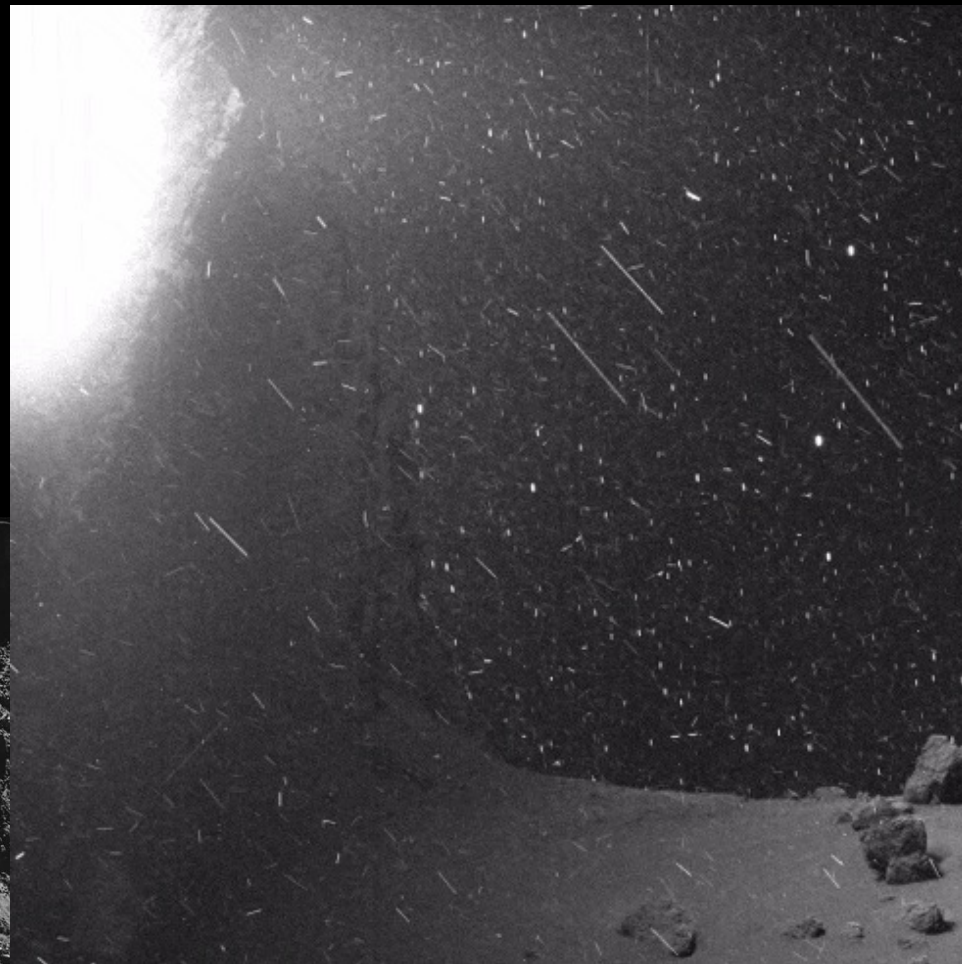
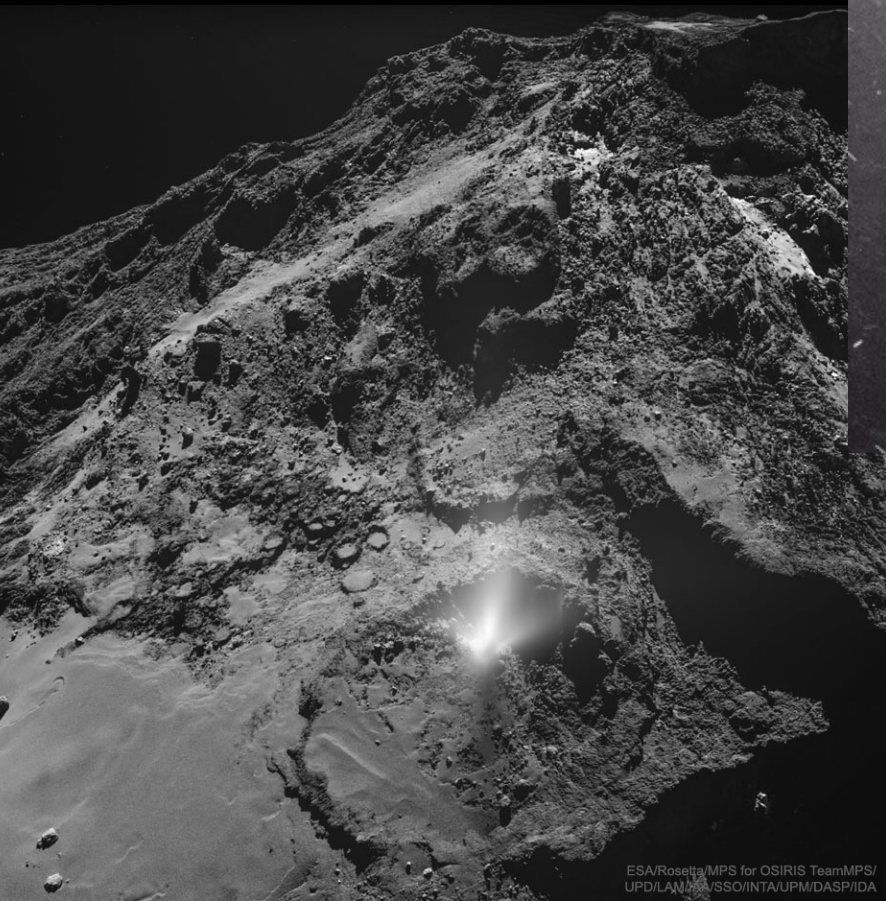


**Some of the mountains
are 11,000ft (3,300m)**

**Mountains made
of water-ice**

**Smooth uncratered terrain
suggests region has been
geologically active in the
last 100 million years**





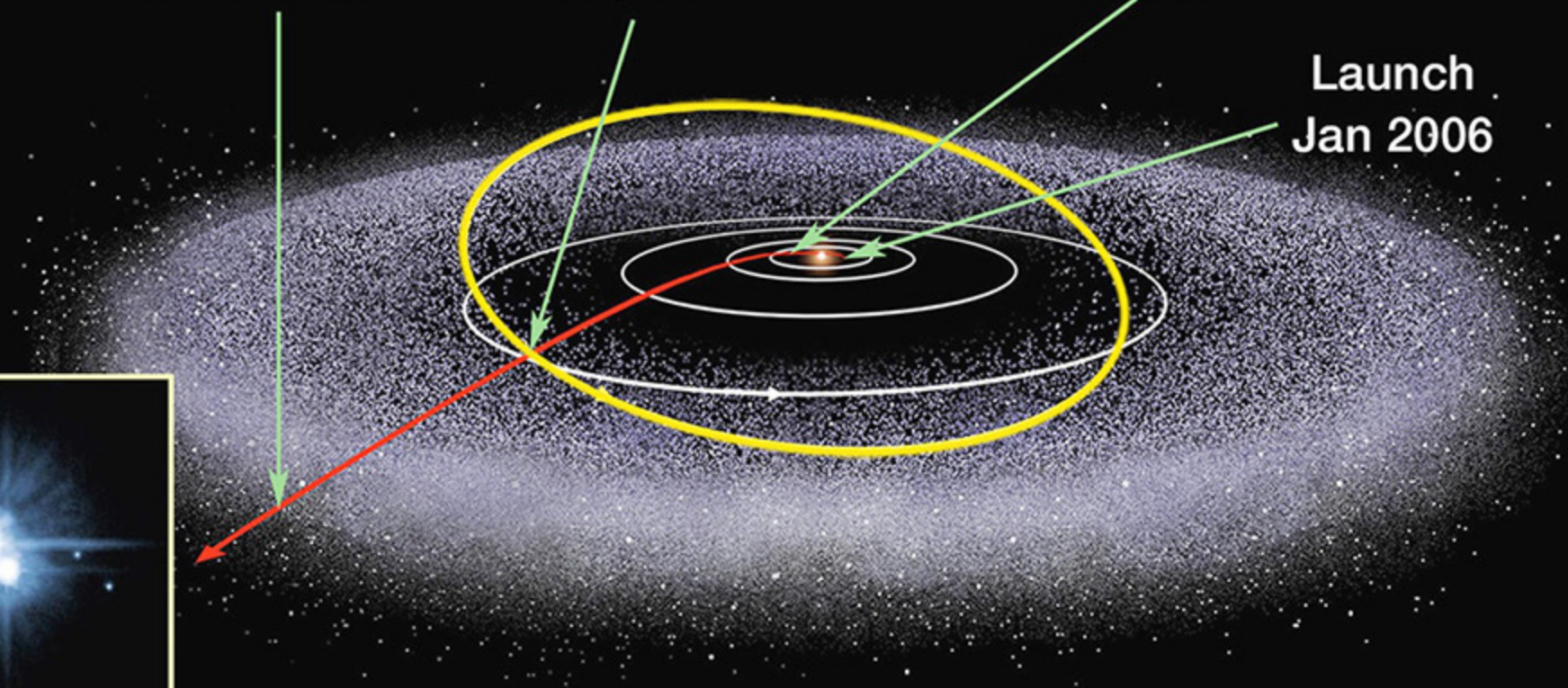
Reconnaissance at the Farthest Frontier

KBOs
2016–2020

Pluto System
July 2015

Jupiter System
Feb 2007

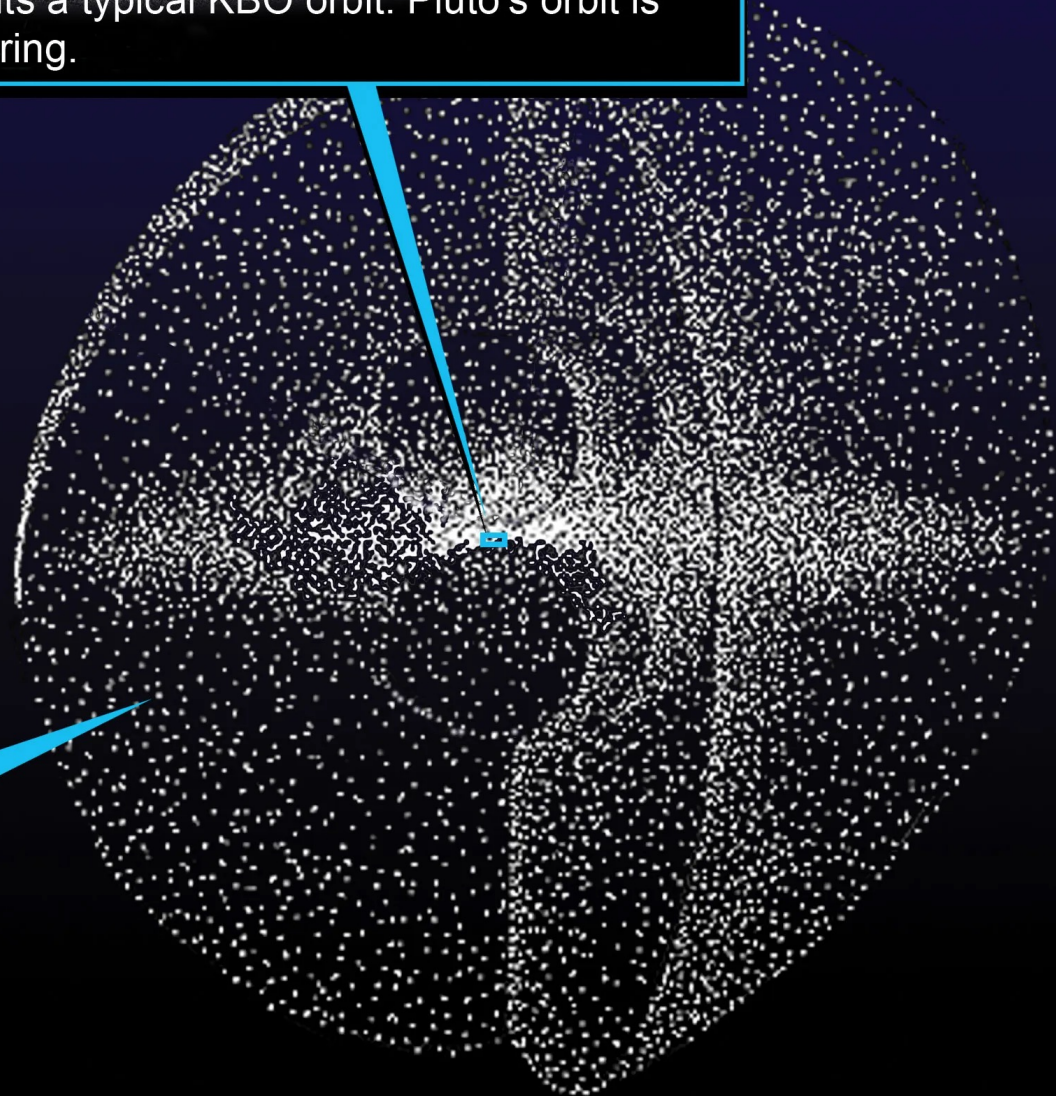
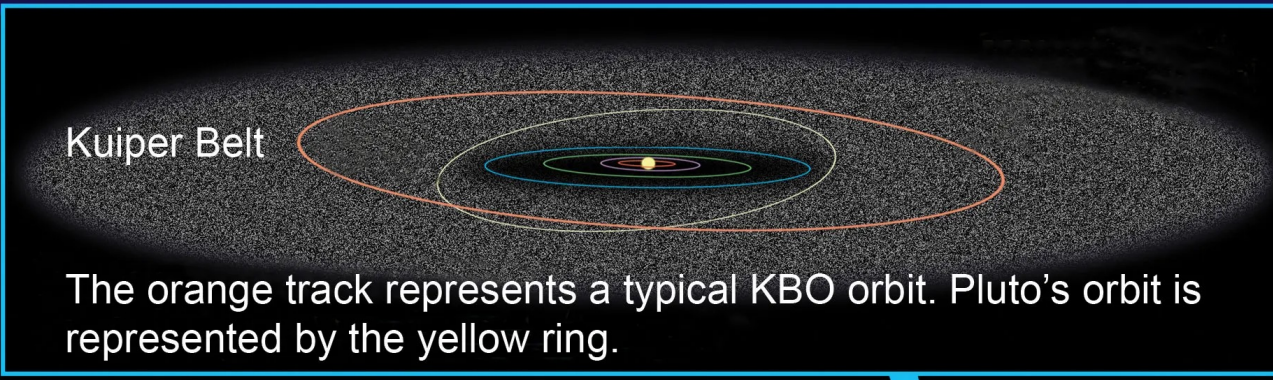
Launch
Jan 2006

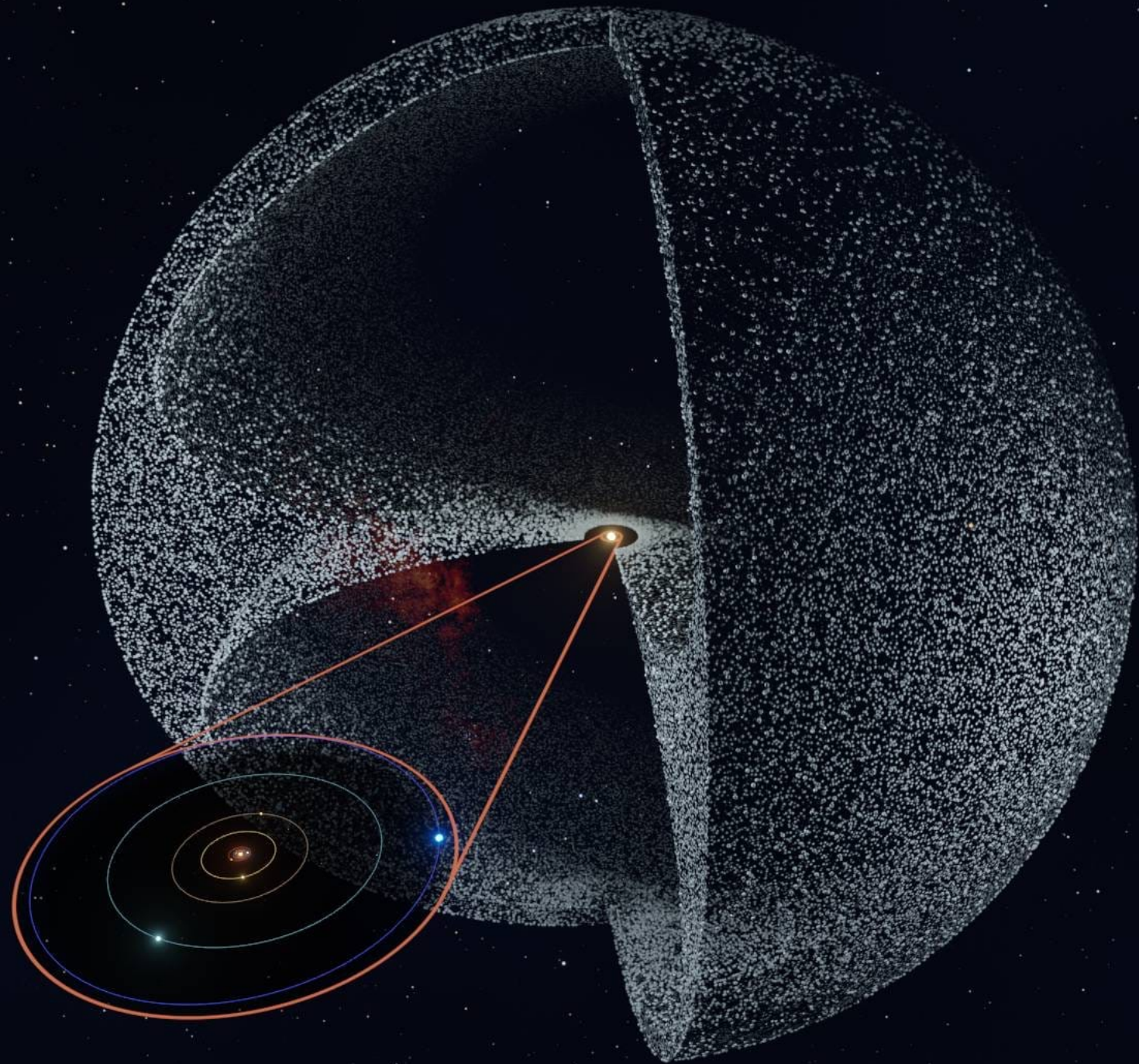


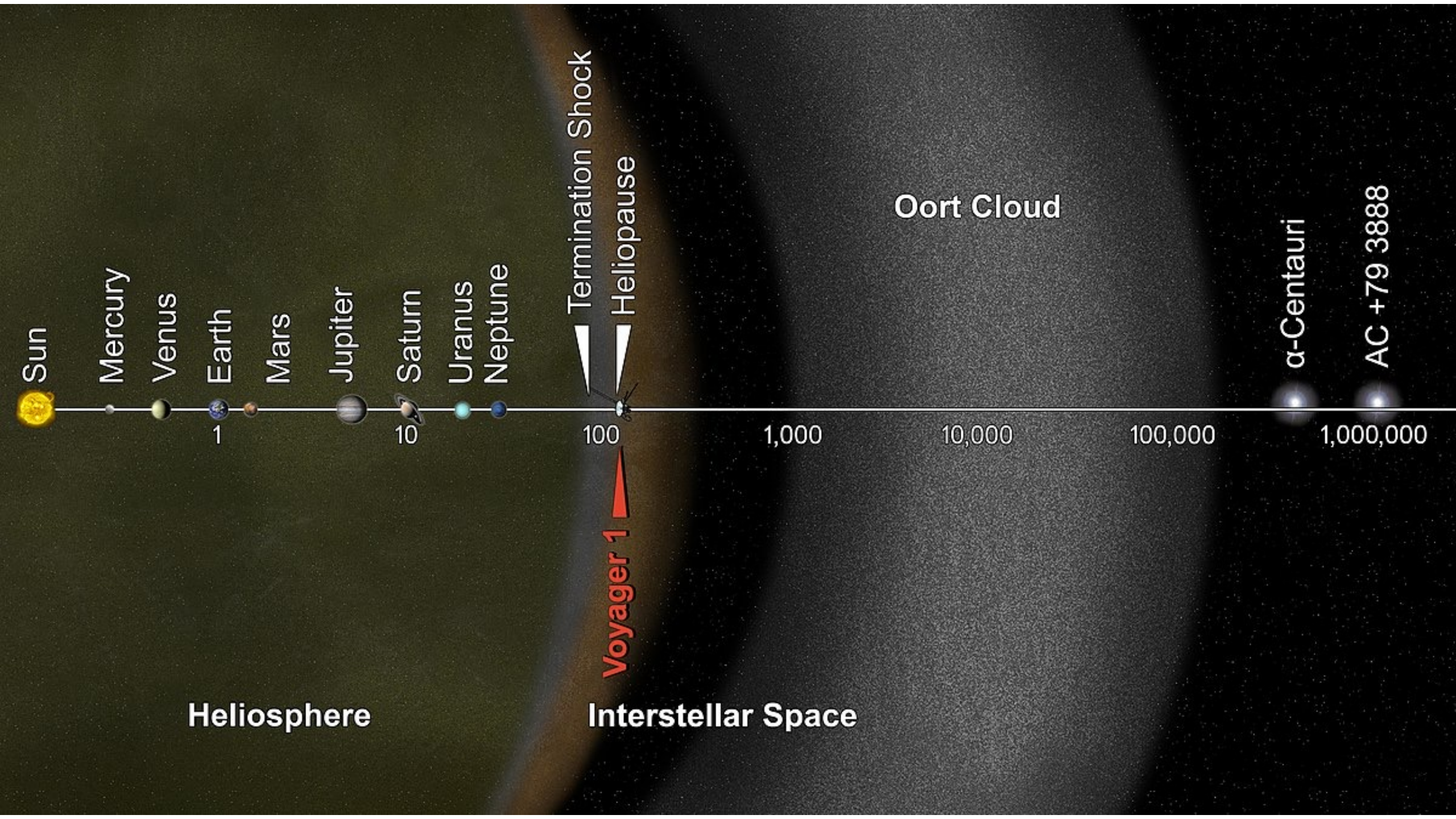
Kuiper Belt

The orange track represents a typical KBO orbit. Pluto's orbit is represented by the yellow ring.

Oort Cloud







Sun

Mercury

Venus

Earth

Mars

Jupiter

Saturn

Uranus

Neptune

Termination Shock

Heliopause

Oort Cloud

alpha-Centauri

AC +79 3888

Heliosphere

Interstellar Space

1

10

100

1,000

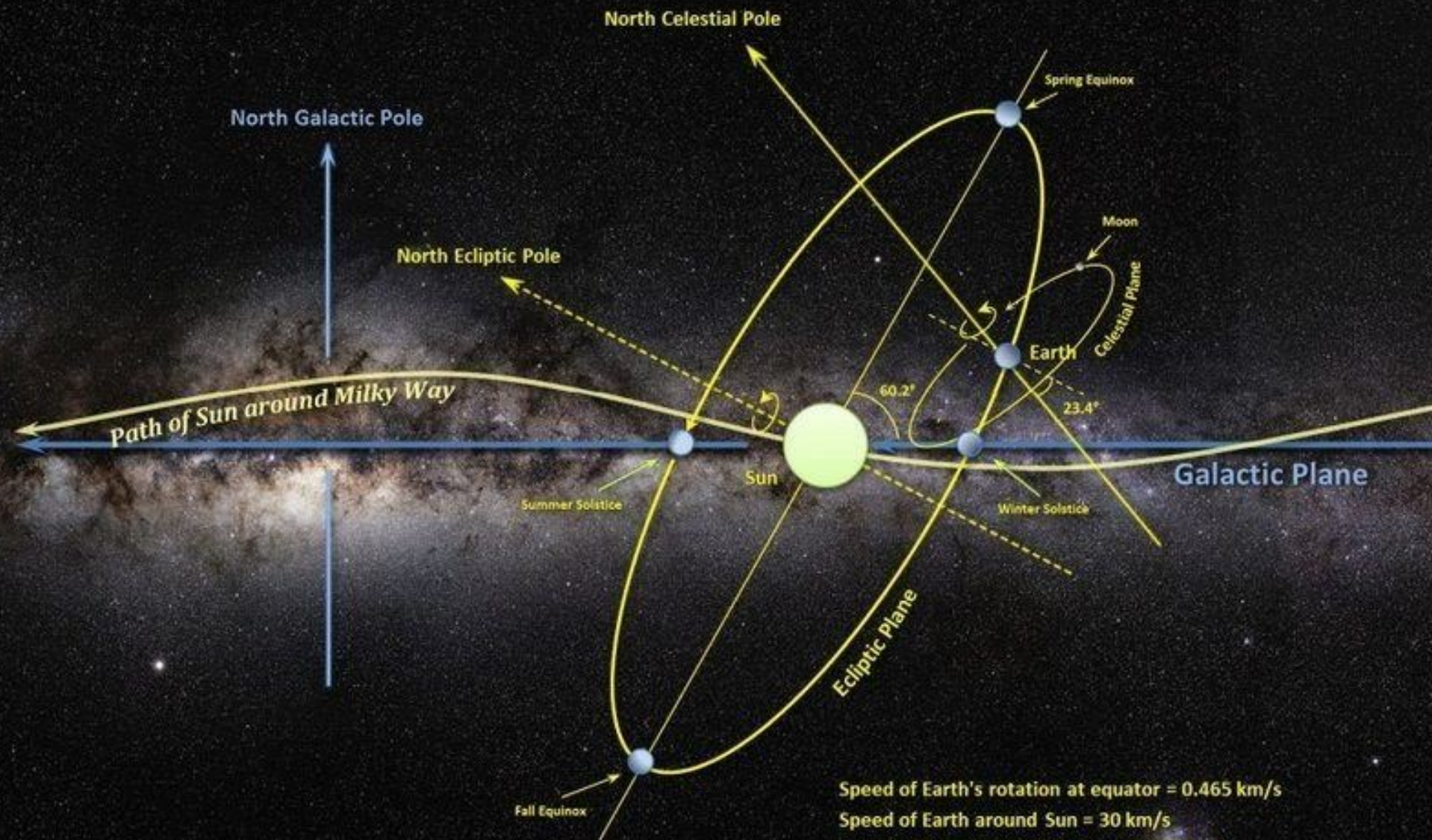
10,000

100,000

1,000,000

Voyager 1

MOTION OF EARTH AND SUN AROUND THE MILKY WAY



Speed of Earth's rotation at equator = 0.465 km/s
Speed of Earth around Sun = 30 km/s
Speed of Sun around Milky Way = 230 km/s
Sun is approximately 26,000 light years from Galactic Center

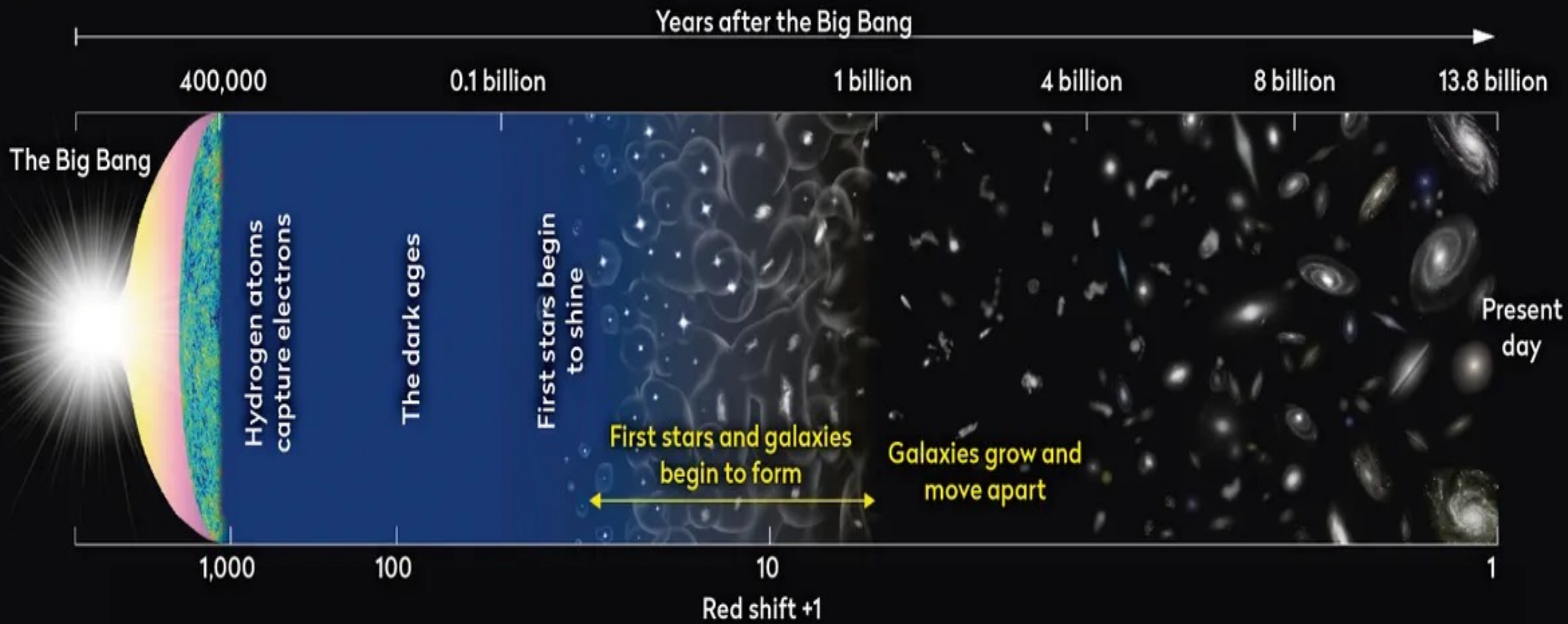
Diagram Not to Scale



Origins

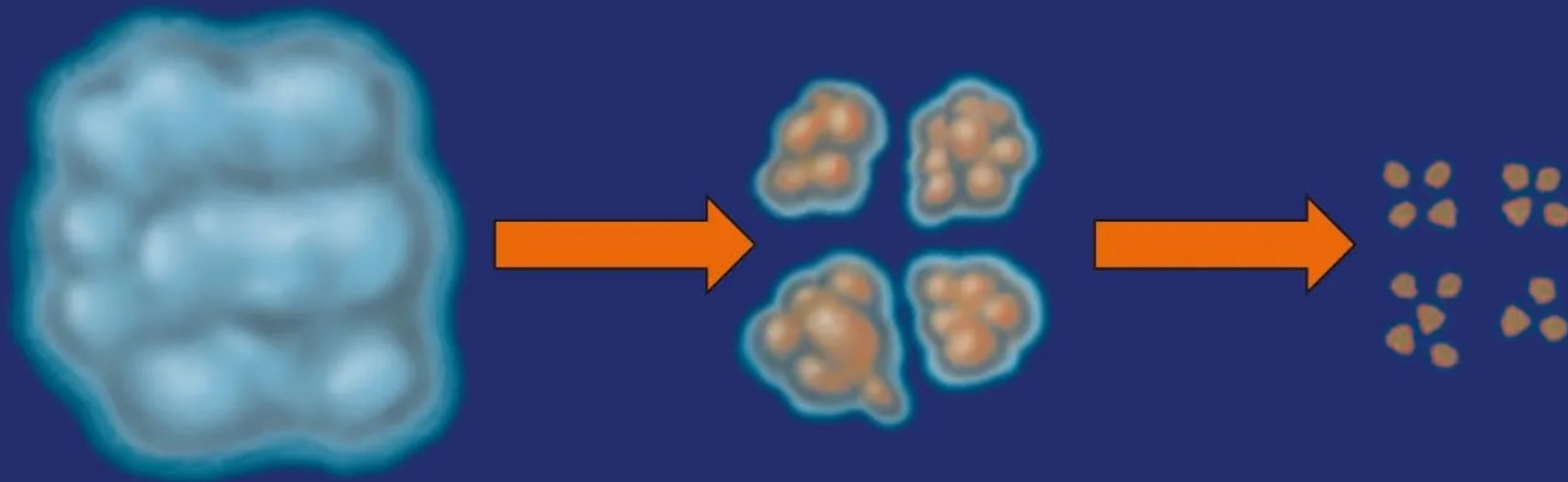
Whoosh!

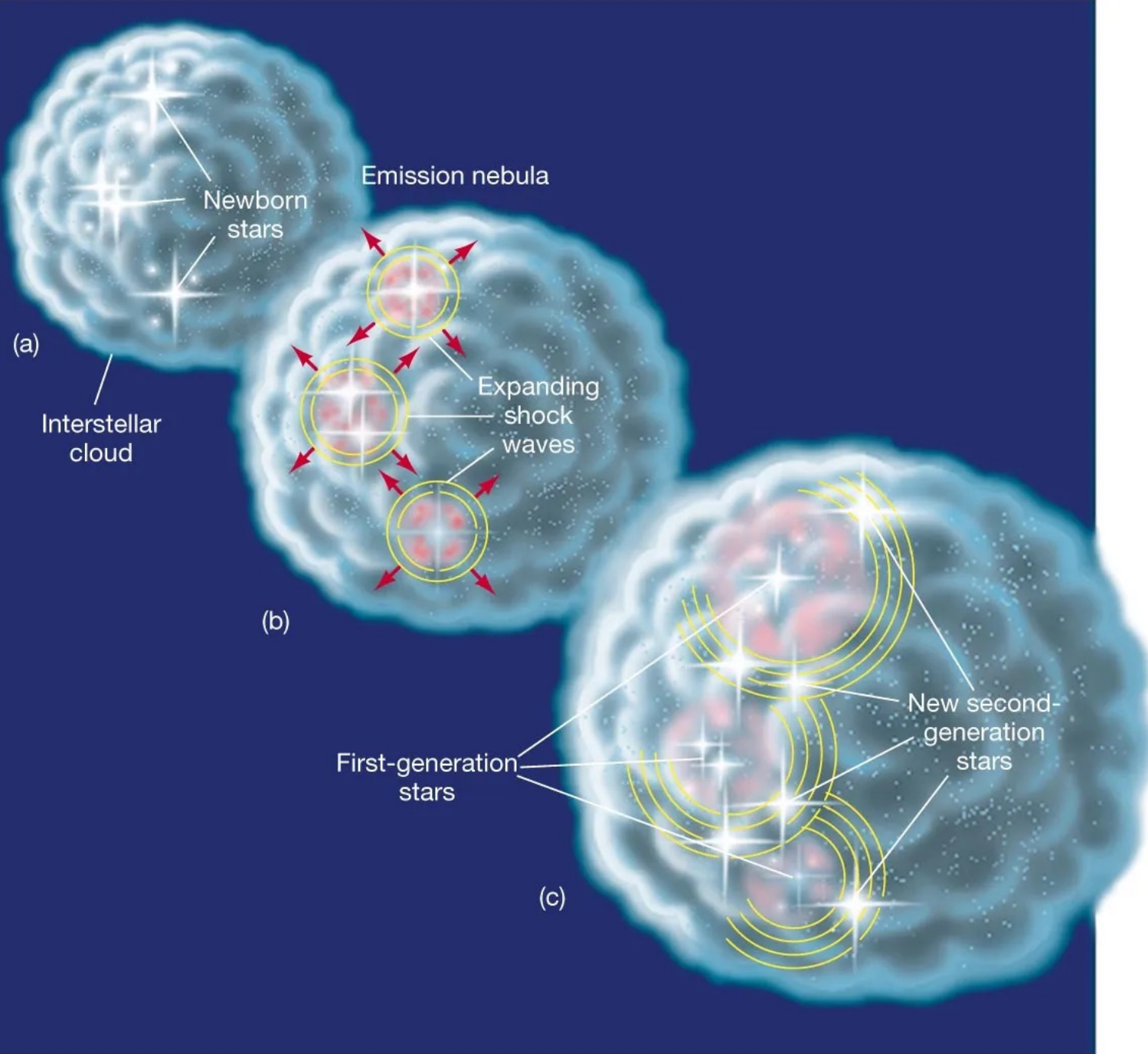
A vibrant blue starburst effect with a bright white and yellow center. Numerous blue and white streaks radiate outwards from the center, creating a sense of motion and energy. The background is a deep blue with scattered white and blue specks, resembling a starry night sky or a digital space environment. The word "Whoosh!" is written in a bold, yellow, sans-serif font at the top center of the image.



Lives of the Stars

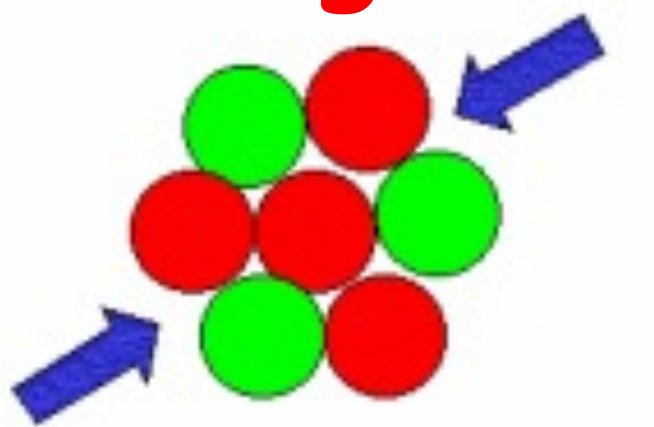
Gravitational Collapse of Stardust Clouds



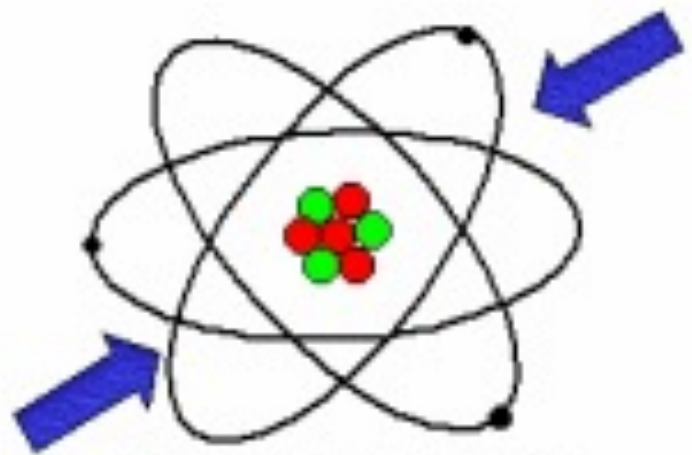


**All Stars
Are
Born in
Clusters**

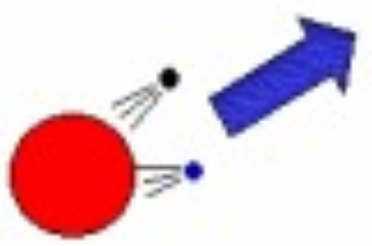
Only Four Forces



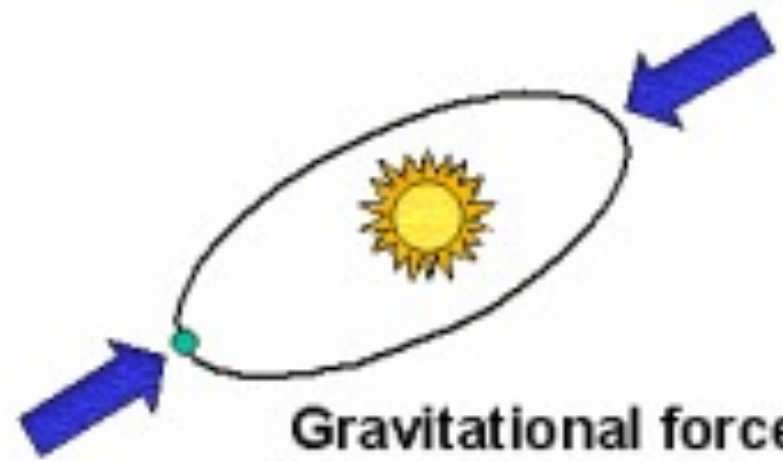
Strong force binds the nucleus



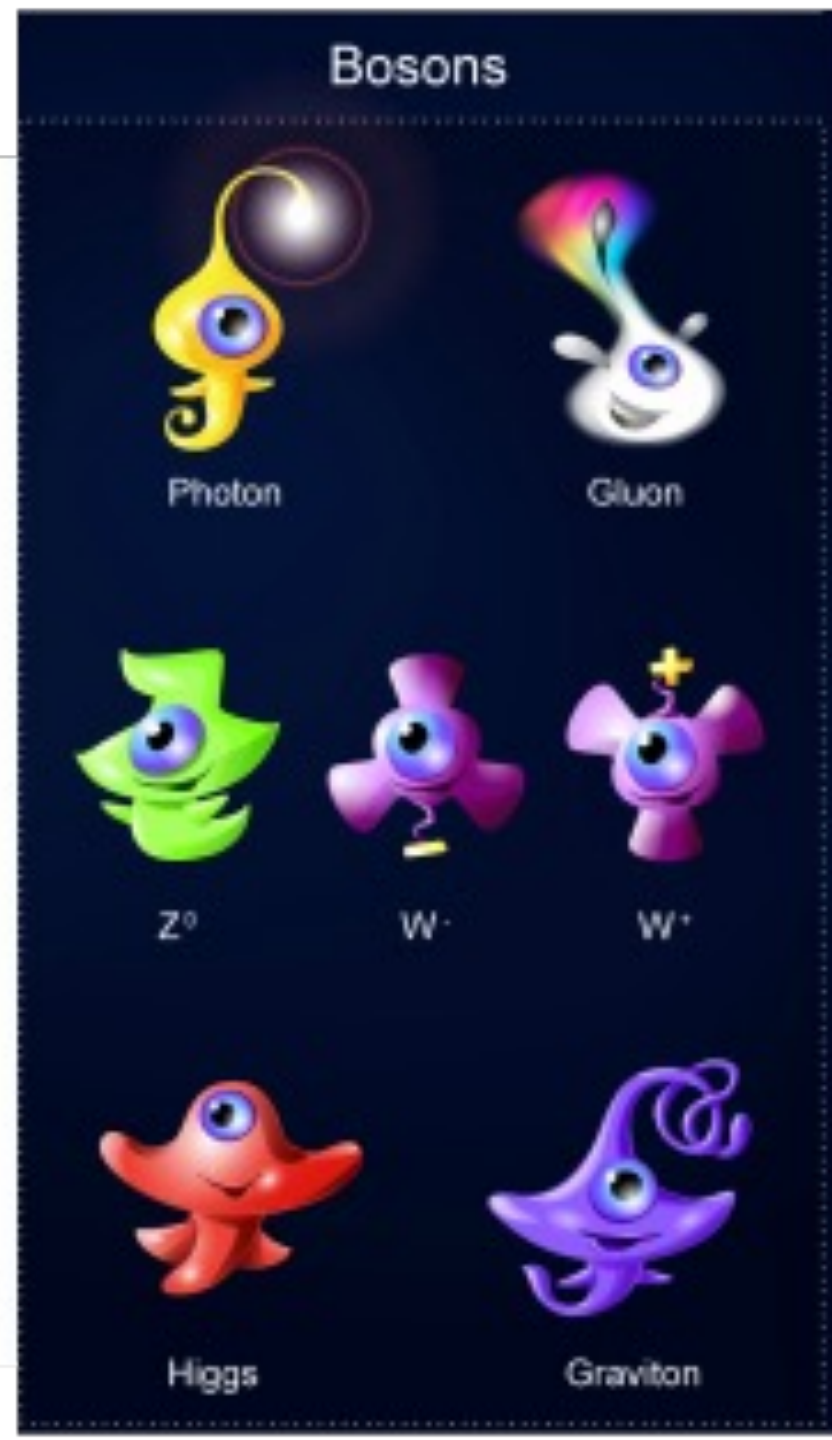
Electromagnetic force binds atoms



Weak force in radioactive decay

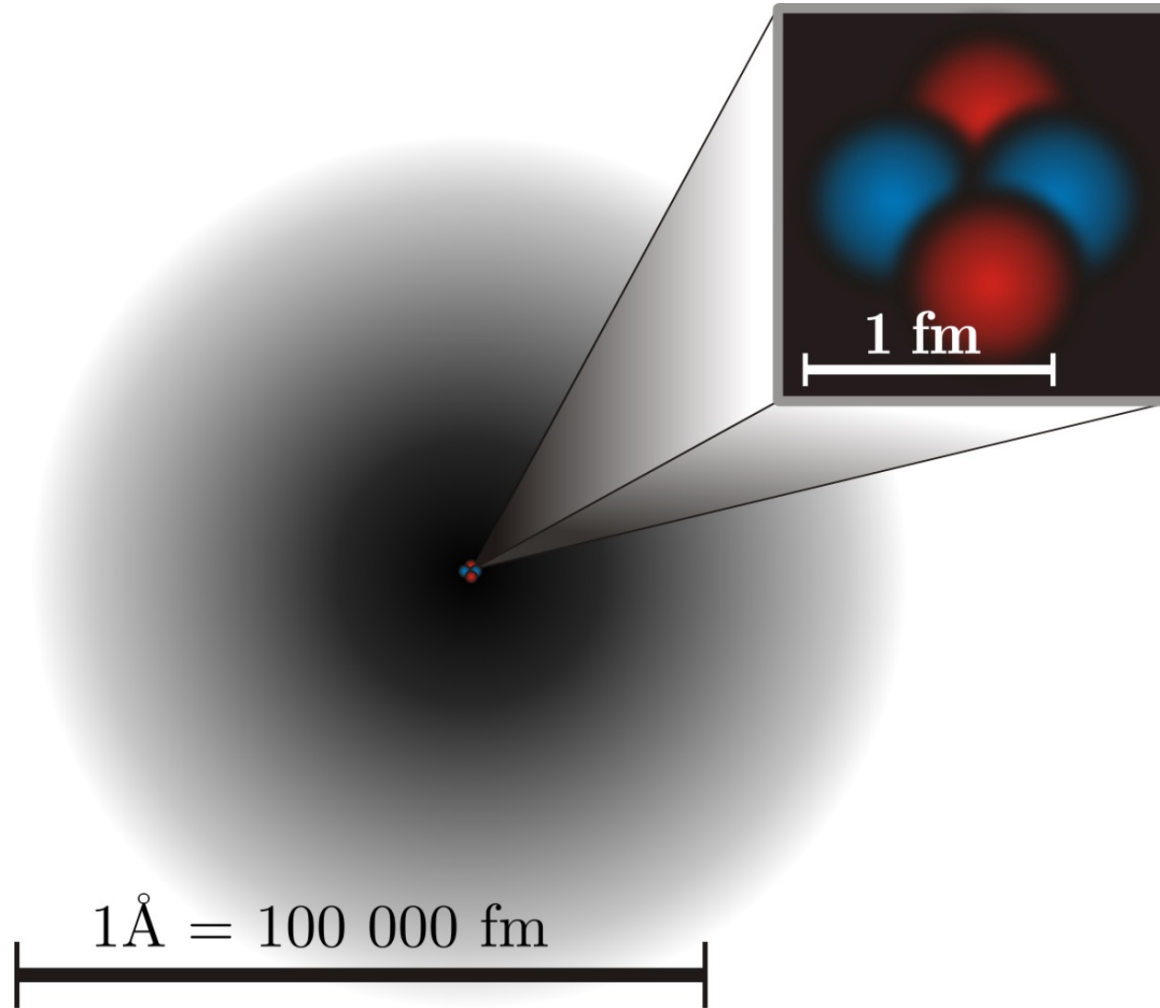


Gravitational force binds the solar system



Atoms are Empty

Although, by volume, an atom is mostly empty space, dominated by the electron cloud, the dense atomic nucleus, responsible for only 1 part in 10^{15} of an atom's volume, contains ~99.95% of an atom's mass. Reactions between internal components of a nucleus can be more precise and occur on shorter timescales, as well as at different energies, than transitions restricted to an atom's electrons.



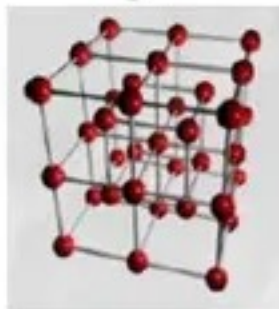
Nuclear scale

Matter



Macroscopic

Crystal



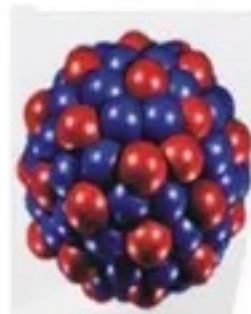
10^{-9} m

Atom



10^{-10} m

Atomic nucleus



10^{-14} m

Nucleon



10^{-15} m

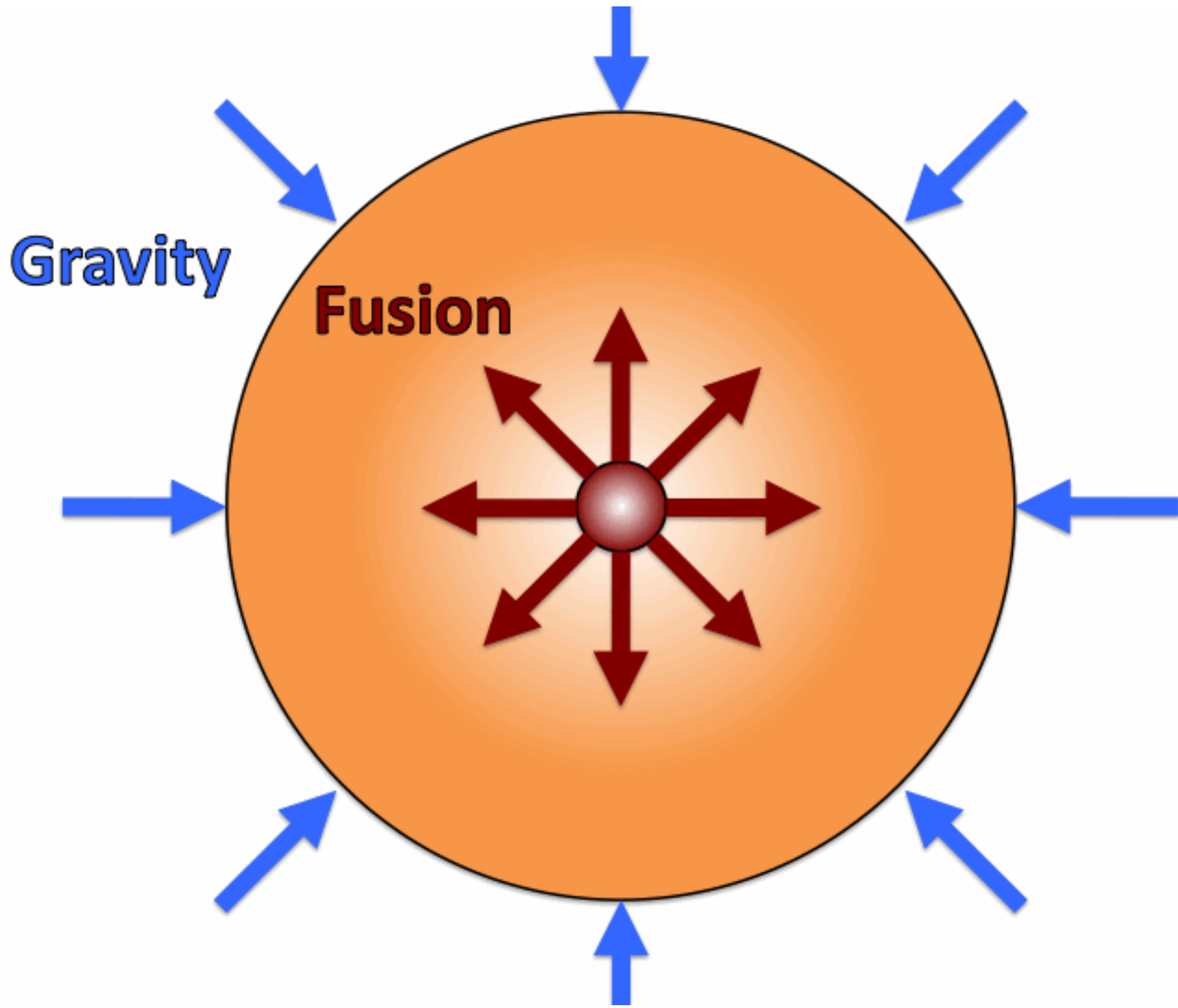
Quark



$< 10^{-18}$ m

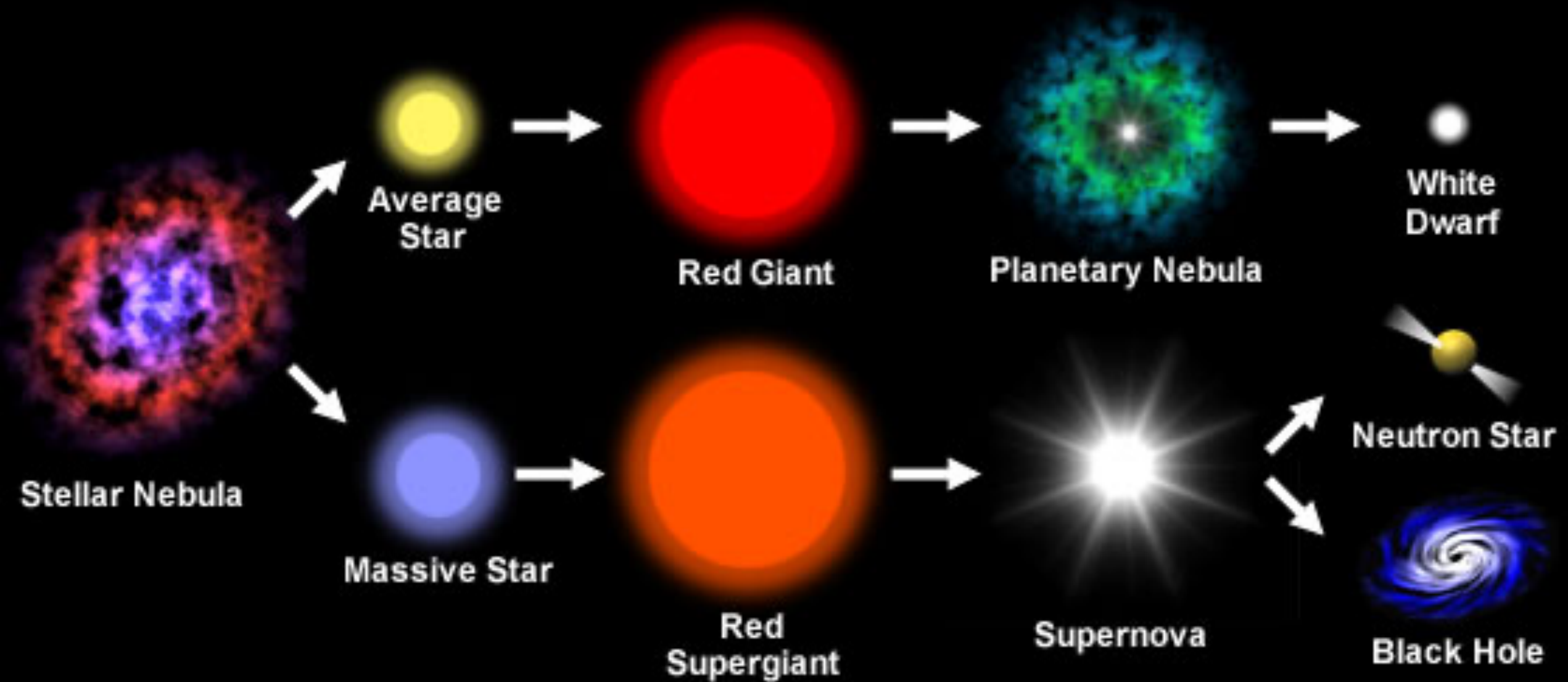
Nuclear physics:

studies the properties of nuclei and the interactions inside and between them

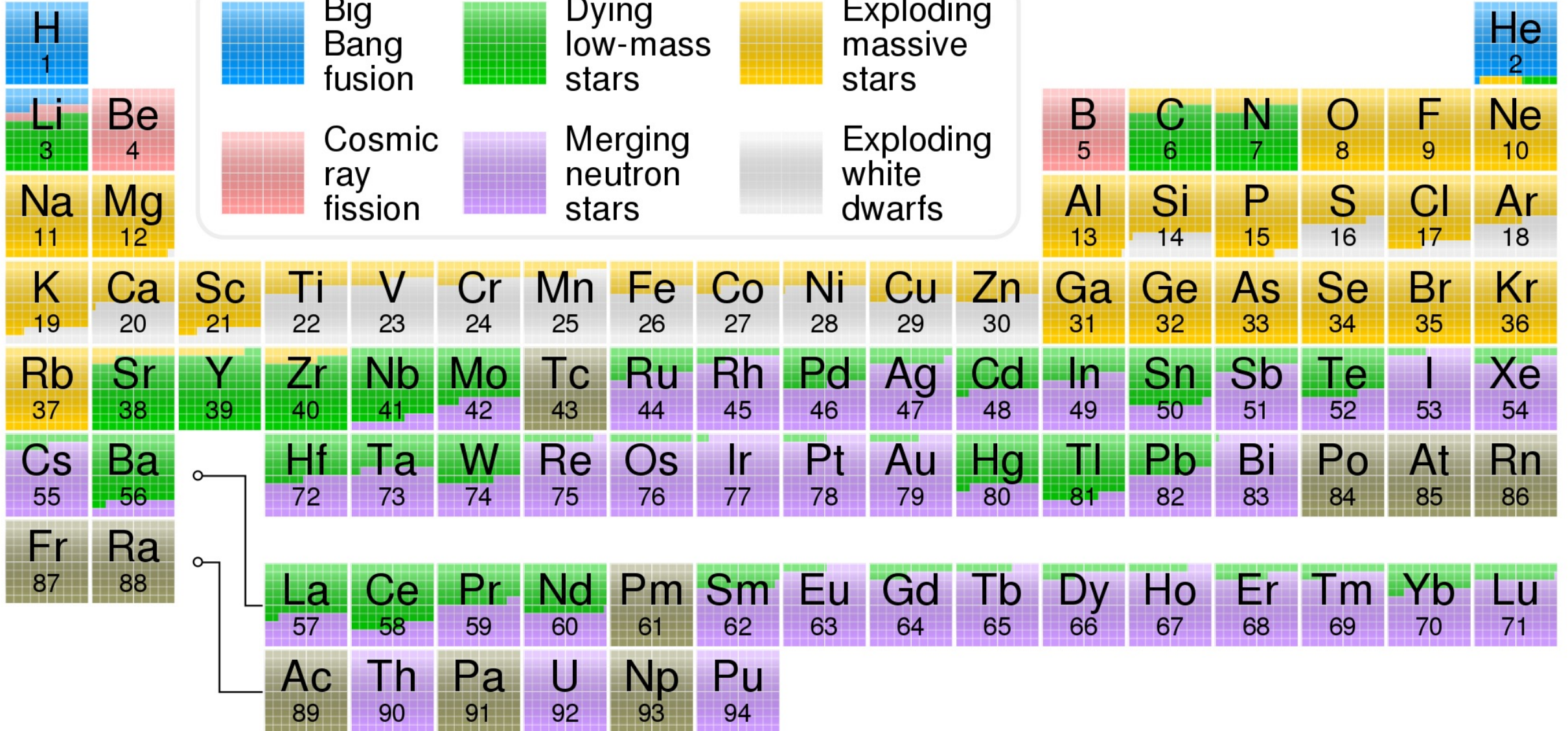


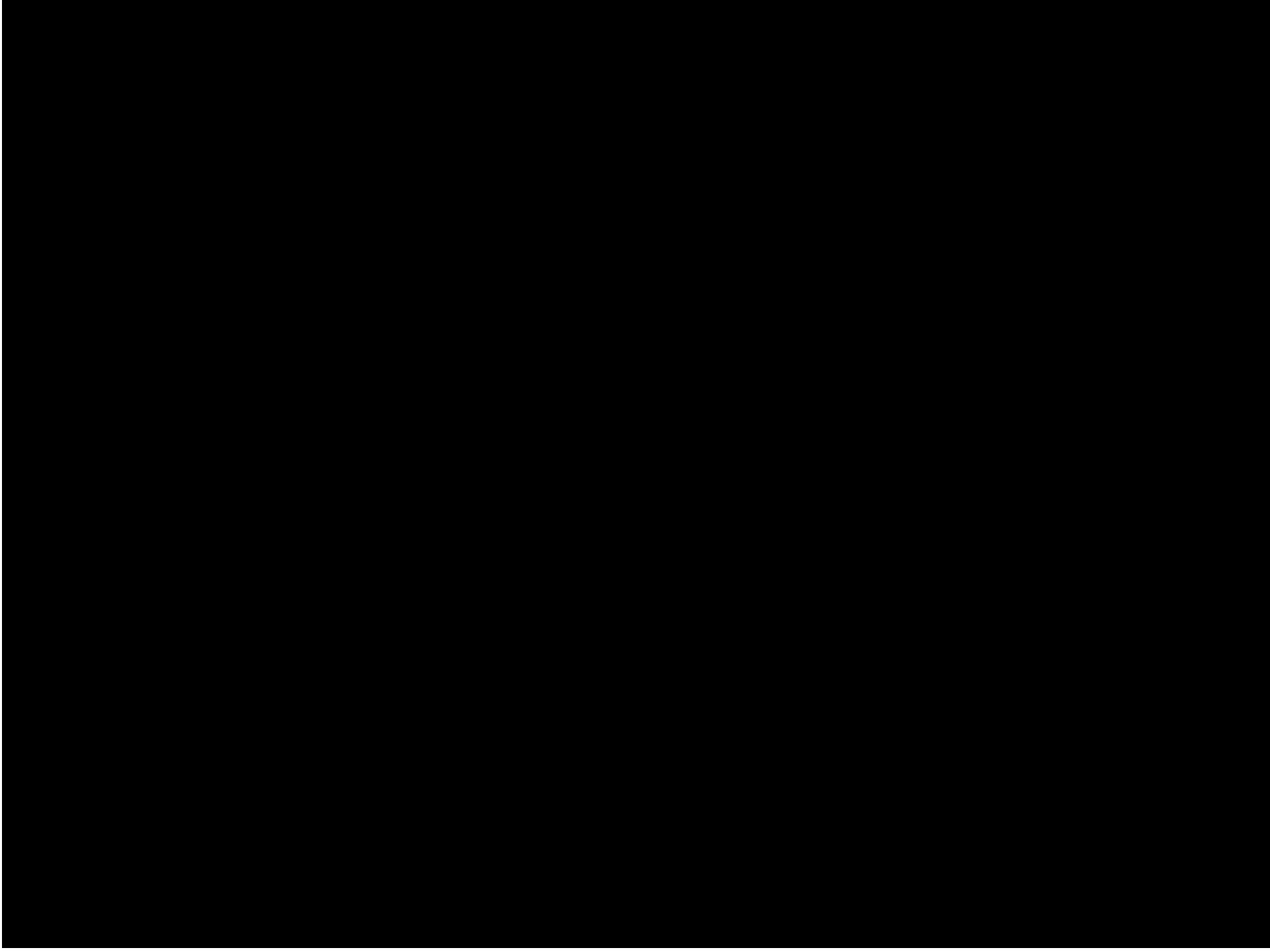
**What
Holds
a Star
Up?**

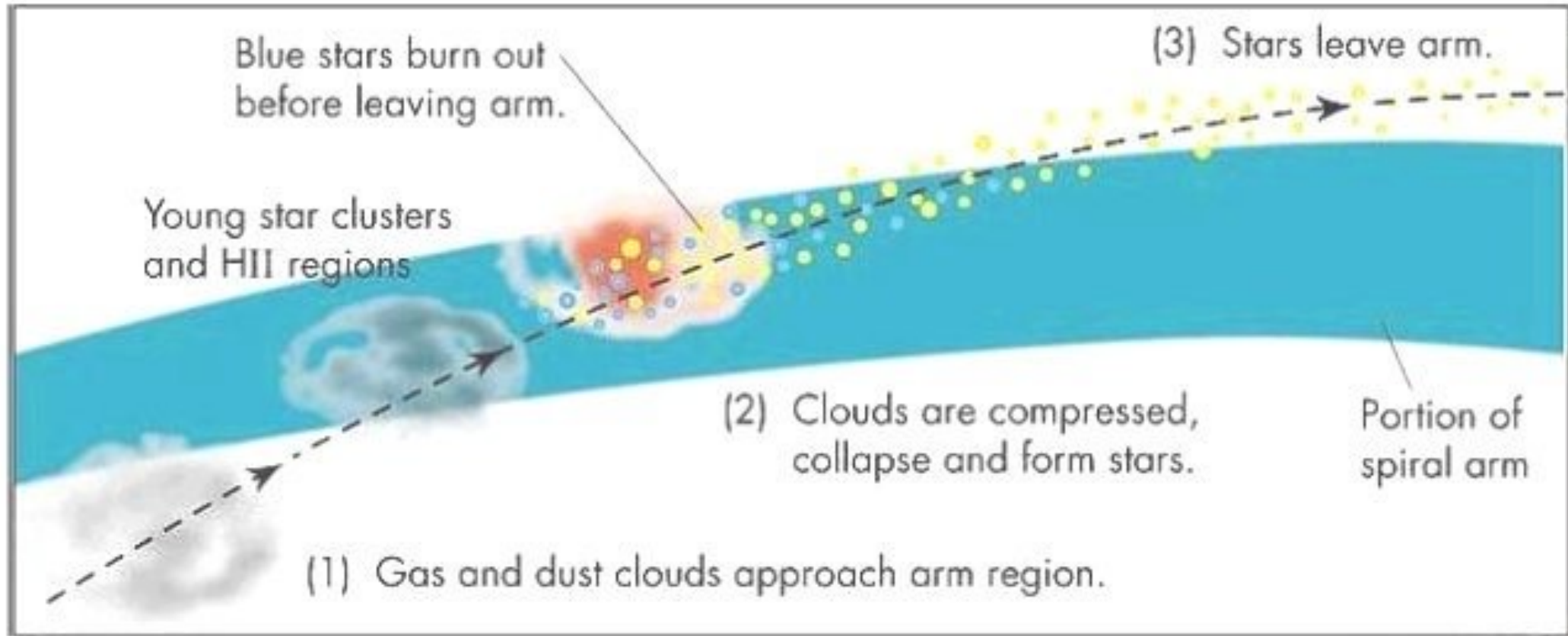
Life Cycle of a Star



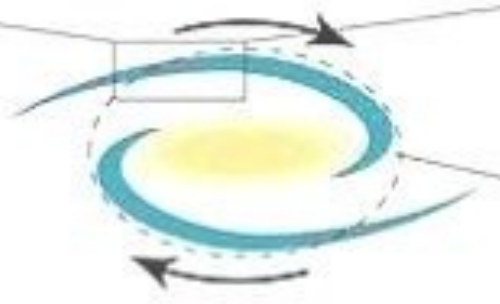
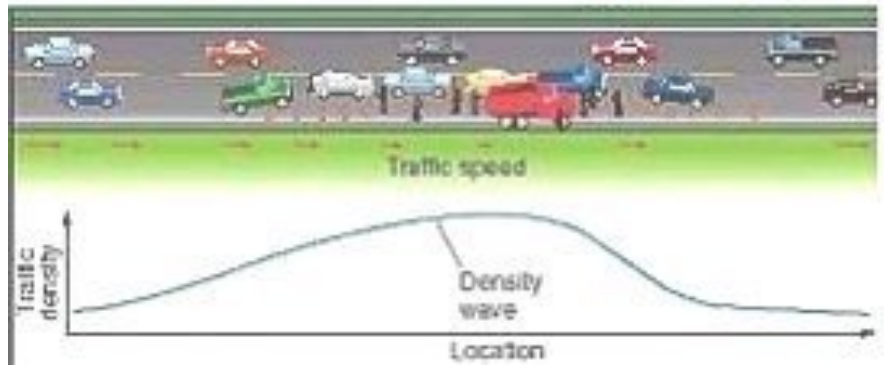
Where ya from?







Traffic Jam Analogy

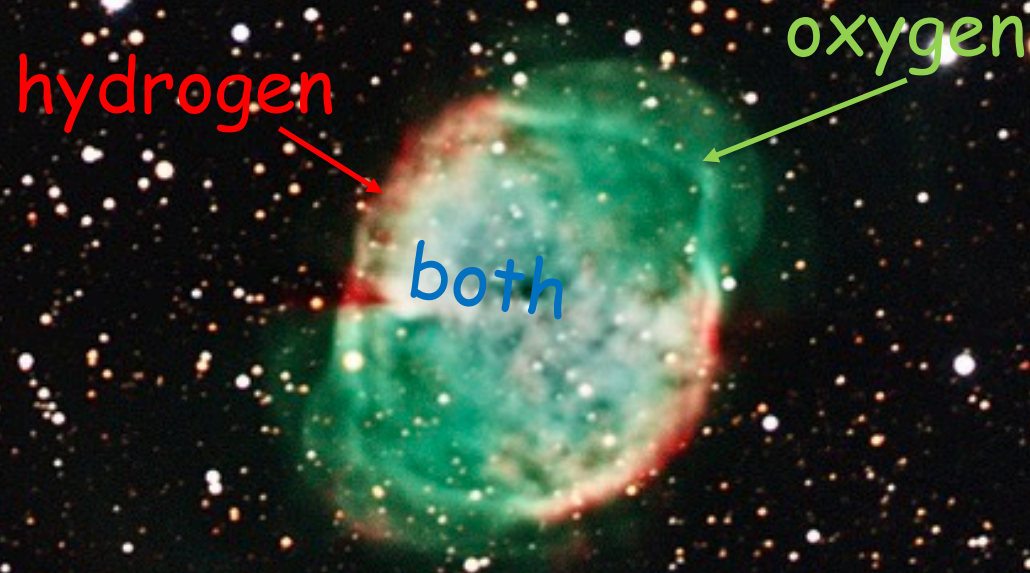


Star and gas orbit

Betelgeuse 2019-2020



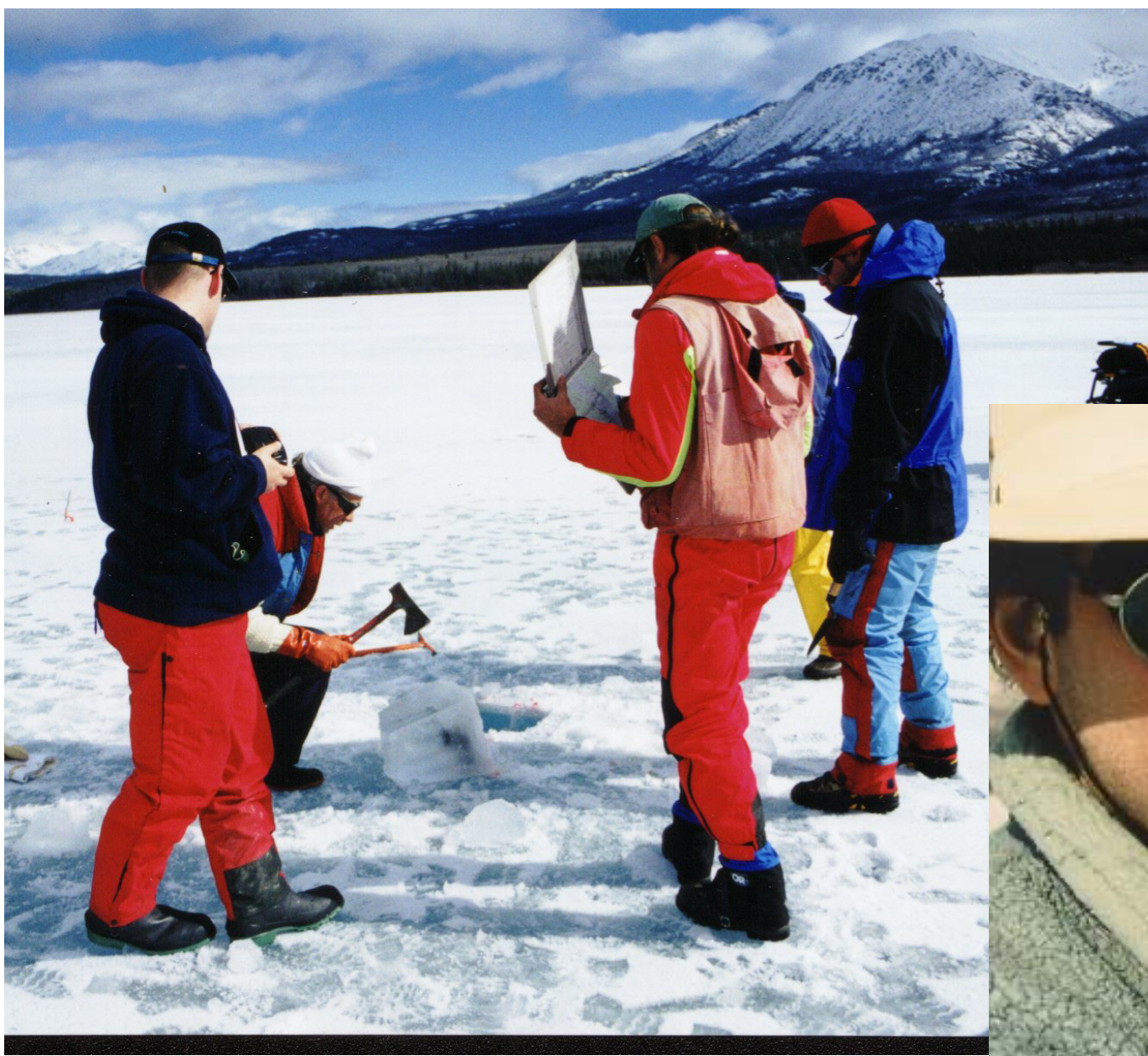
Where Ya From?



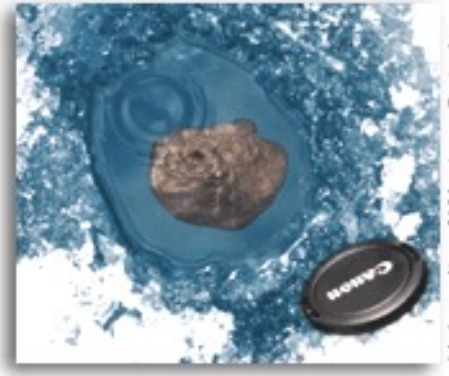
Wisps of alchemy blown on stellar winds

all astro-images by Scott Denning





Tagish Lake meteorite



University of Western Ontario,
University of Calgary



(University of Western Ontario,





Discovery: Cosmic Dust Contains Organic Matter from Stars

By Denise Chow October 26, 2011



Cite this: *Phys. Chem. Chem. Phys.*, 2014, 16, 3381



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Physics of Life Reviews 8 (2011) 307–330

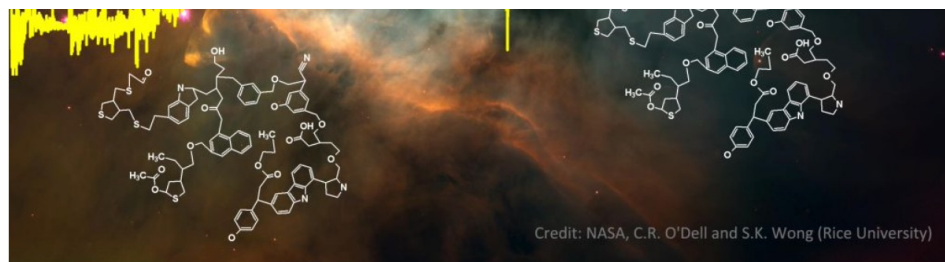
PHYSICS of LIFE
reviews

www.elsevier.com/locate/plrev

Review

Photochirogenesis: Photochemical models on the absolute asymmetric formation of amino acids in interstellar space

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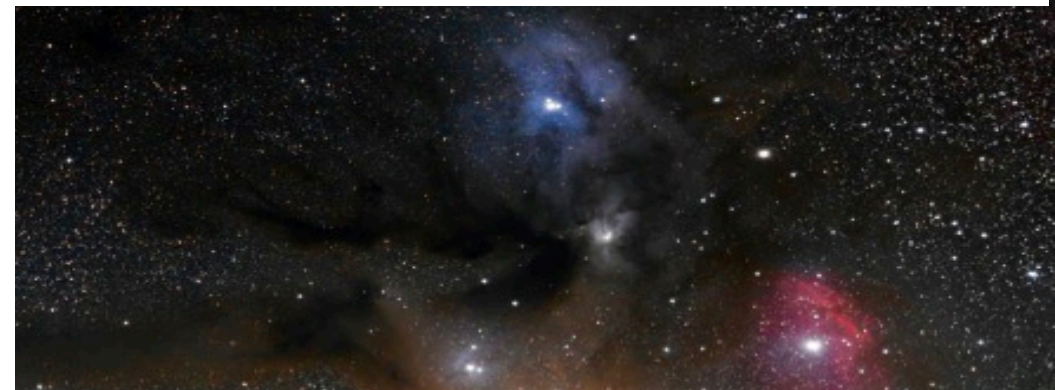


Credit: NASA, C.R. O'Dell and S.K. Wong (Rice University)

A spectrum from the European Space Agency's Infrared Space Observatory superimposed on an image of the Orion nebula, where these complex organics are found. (Image: © NASA, C.R. O'Dell, S.K. Wong (Rice University))

Hydrogenation of PAH molecules through interaction with hydrogenated carbonaceous grains

John D. Thrower^{*}, Emil E. Friis, Anders L. Skov, Bjarke Jørgensen and Liv Hornekær^{*}



Researchers Find Extraterrestrial Protein in Meteorite Acfer 086

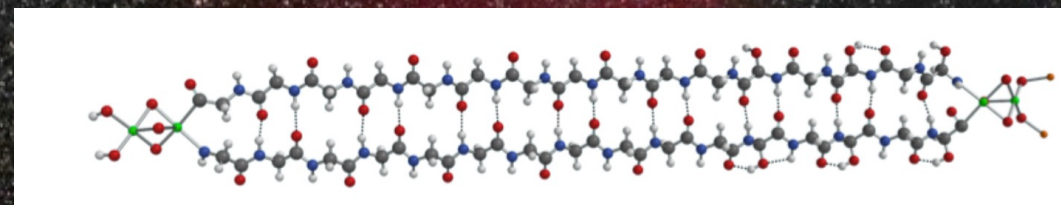
Mar 26, 2020 by Sergio Prostak

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Published in

Biology
Chemistry

A research team led by Harvard University scientist **Julie McGeoch** has found a never-before-seen protein inside a meteorite called Acfer 086.



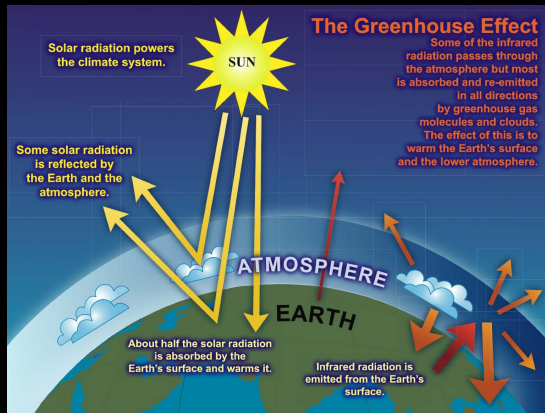
Selective Transparency

Glass, Air

- Transparent to visible light
- Opaque to IR

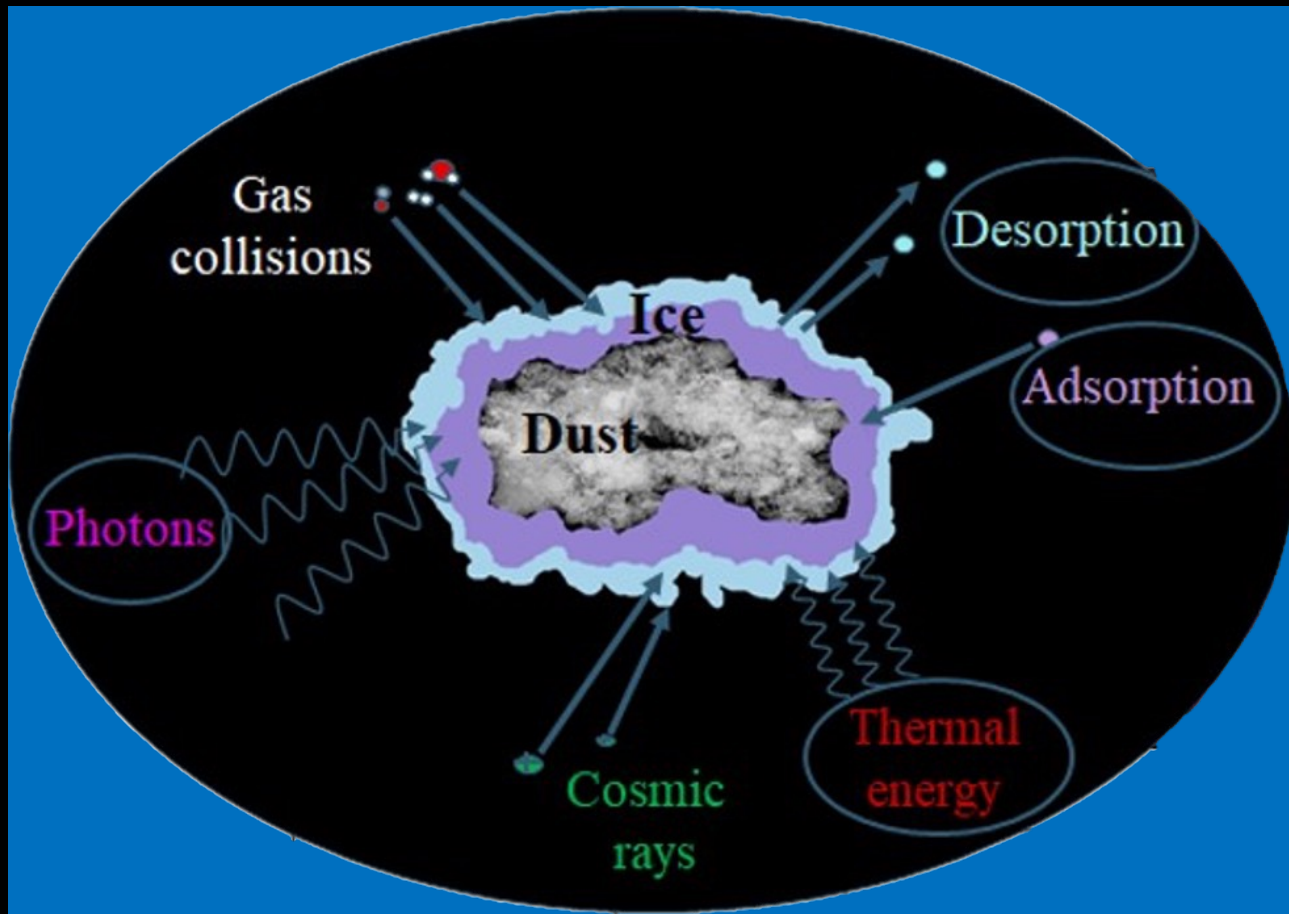
Cosmic Dust

- Transparent to IR
- Opaque to UV & Vis



Gas & dust cool by emitting IR to space
Outer layer scatters & absorbs UV/Vis, protects core
Cryogenic cooling (10 K!) allows chemistry, collapse

Cosmic Ice & Dust

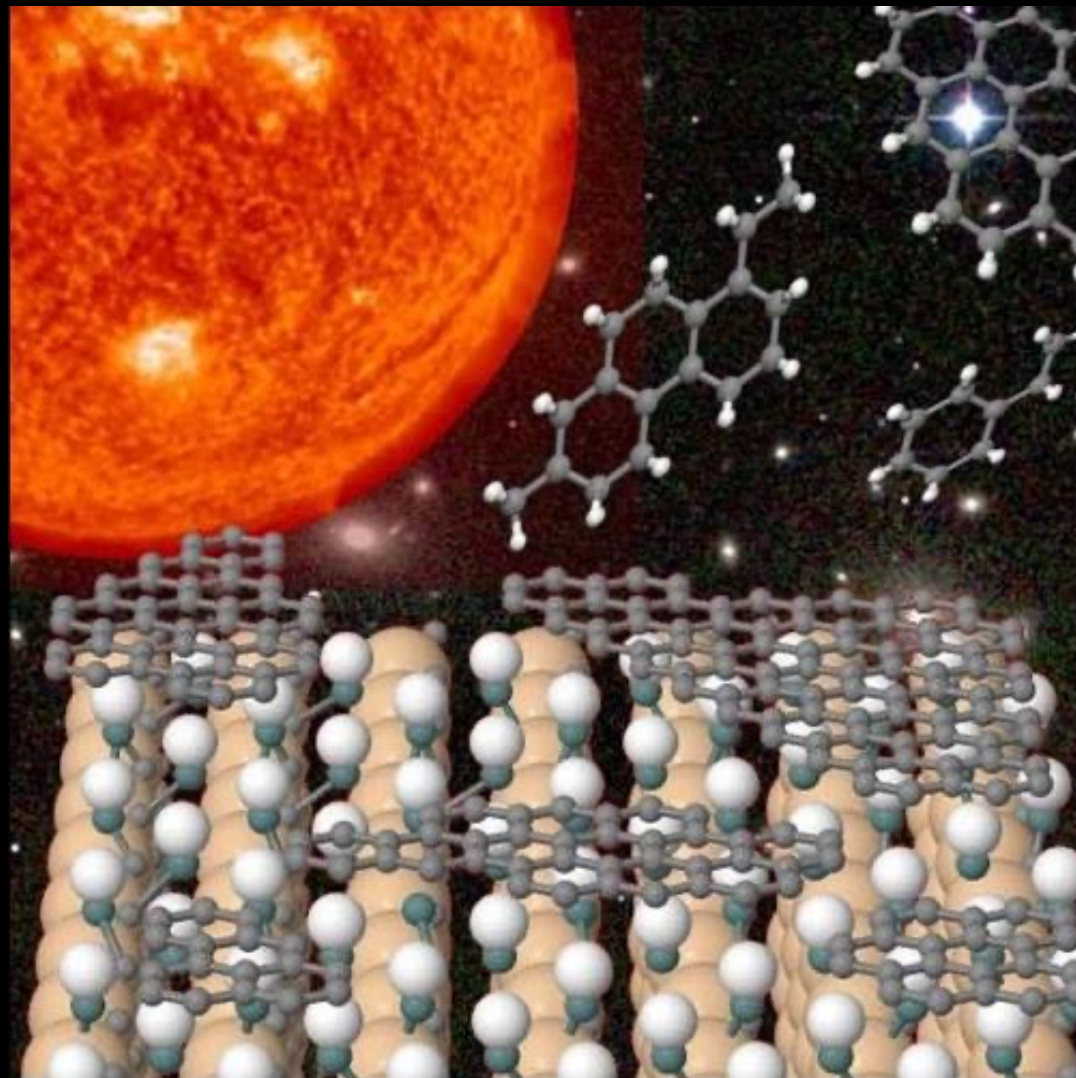
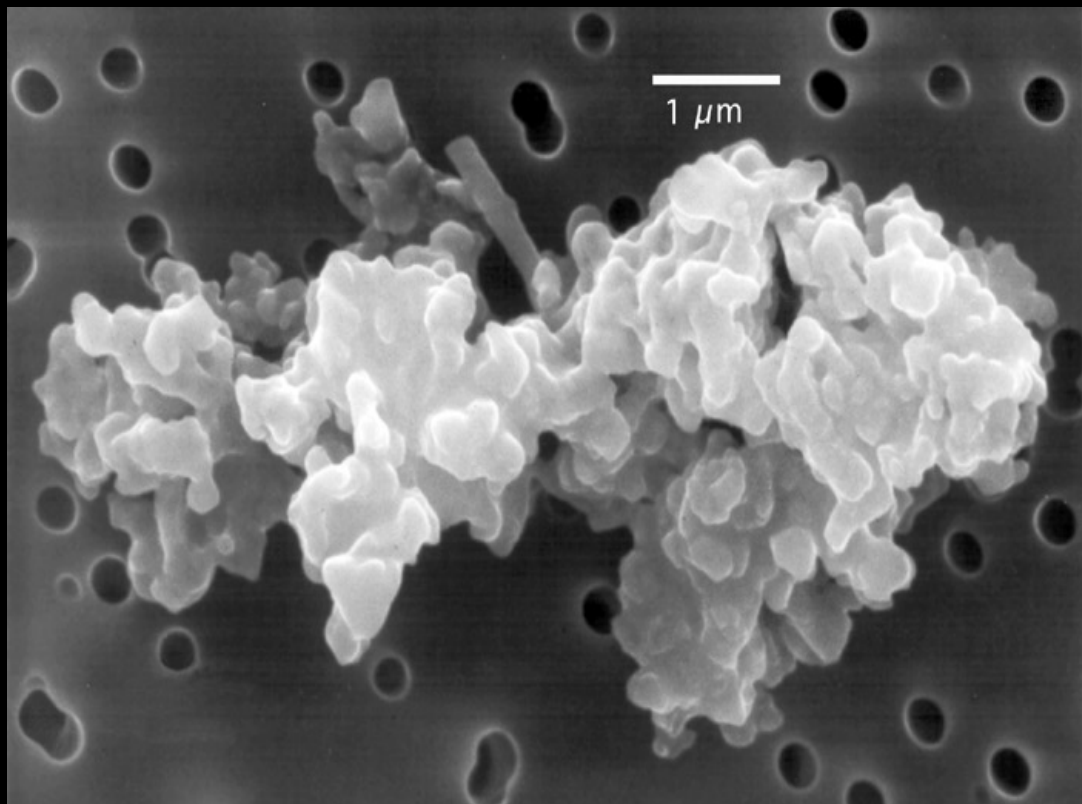


H₂O, CO, CO₂, CH₄, NH₃, and CH₃OH

- Mixed-phase chemistry
- Physical protection
- Surface interactions
- *Radiation environment*

Prebiotic Volatiles in the Cryogenic Dark

Star Smoke

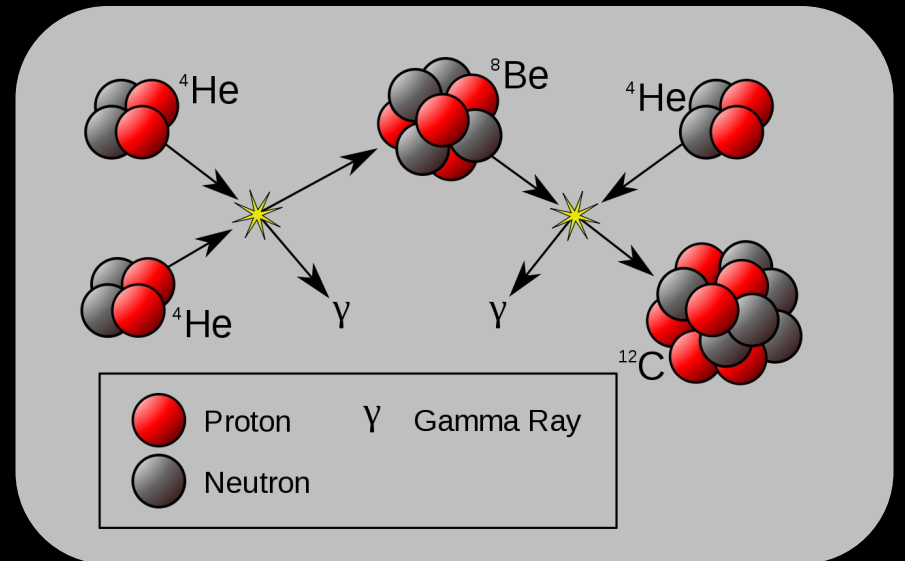
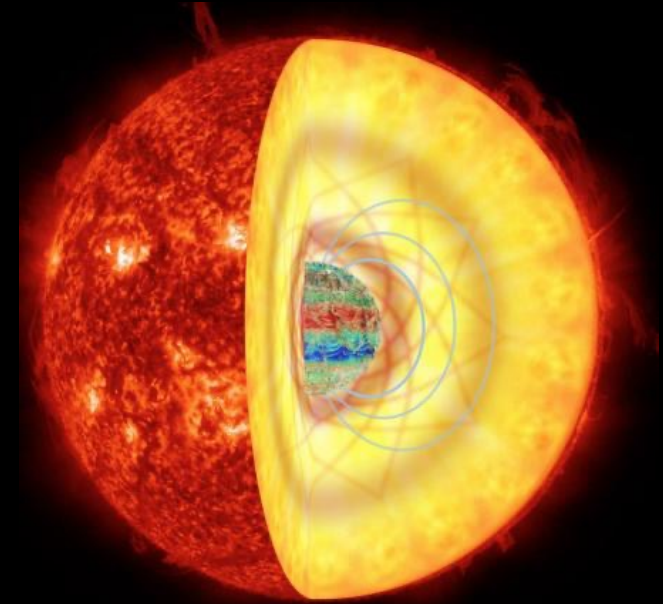


What is Carbon?

- Element #6 in the Periodic Table
 - 6 protons
 - 6 neutrons (usually)
 - 6 electrons
- All the carbon in the universe was made inside of stars!
 - 2 Hydrogen atoms make a Helium atom
 - 3 Helium atoms make a Carbon atom

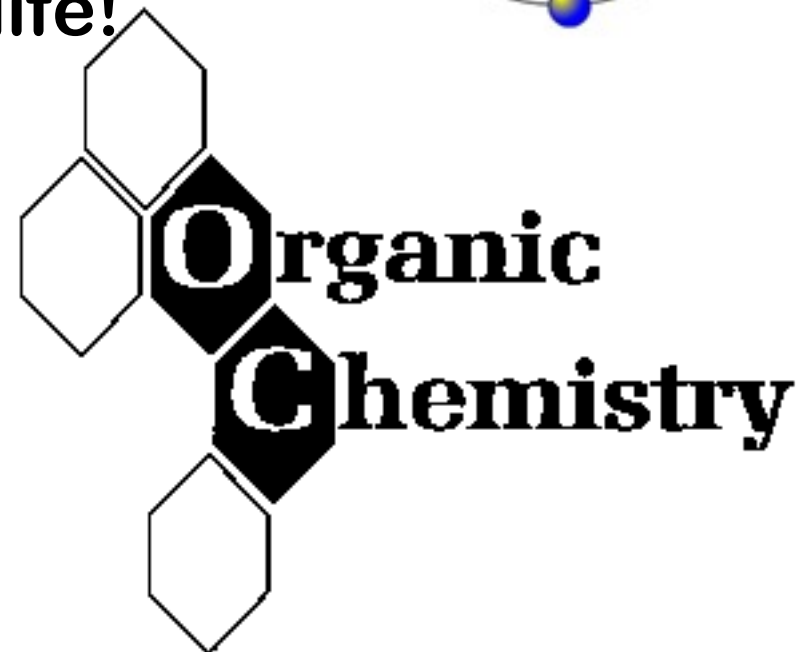
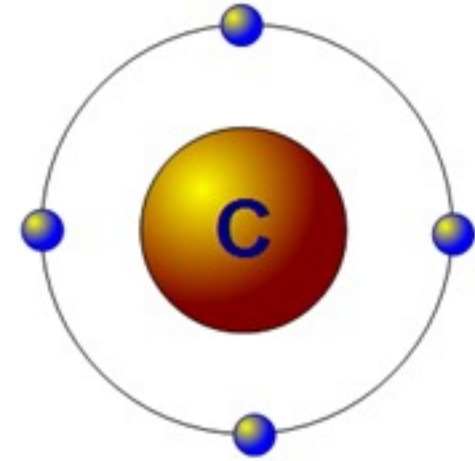
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					2
					He
					4.0026
3A	4A	5A	6A	7A	
5	6	7	8	9	10
B	C	N	O	F	Ne
10.81	12.01115	14.0067	15.9994	18.9984	20.179
13	14	15	16	17	18
Al	Si	P	S	Cl	Ar
26.9815	28.0855	30.9738	32.06	35.453	39.948

6	Atomic number
C	
12.01115	Atomic mass



Carbon is Central

- Outer electron shell can donate or receive 4 “valence” electrons
- Neither positive nor negative –
Chains! Rings! Sheets! Tubes!
- Basis for all of organic chemistry & life!



μ Cephei (“Erakis” ~ “Arrakis”)

Herschel’s
Garnet Star

a carbon
star



Dust to Dust















Dust Factories



photographed from my back yard











Where Do We Come From

- “If you want to make an apple pie from scratch, you must first invent the Universe”
- We’re thoroughly bound up with the entire story of creation:
 - From light to quarks ...
 - to plasma ...
 - to atoms ...
 - to stars ...
 - to dust ...
 - to planets ...
 - to plants ...
 - to people ...

What Are We?

- Descendants of stars ...
- Weavers of recycled stuff ...
- A flame that dances on the surface of matter
- Lucky mud ...
- Current in a circuit flowing from the Sun ...
- Heirs to billions of years of biological evolution ...
- Agents of creation and destruction ...

Where Are We Going?

- To dust we shall return
- Our material and energy are eternal
- We flow through it & it is us &
our stuff will flow on through others
- We are given the grace of life & love
- We have the agency to choose how to
spend our precious time

